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Chemical Bonding Webquest

Visit the following webpage: http://www.chem4kids.com/files/atom_bonds.html

Bonding Basics

1. What do atoms need to be "happy"?
2. When do atoms give up their electrons?
3. When do atoms look for more electrons?
4. Look at sodium.
 - a. How many total electrons does sodium have? _____
 - b. How many filled shells does it have? _____
 - c. How many extra electrons does it have? _____
 - d. How many electrons do you think sodium will give away? _____
5. Look at fluorine.
 - a. How many total electrons does fluorine have? _____
 - b. How many filled shells does it have? _____
 - c. How many missing electrons does it have? _____
 - d. How many electrons do you think fluorine will accept? _____
6. Draw a diagram showing what happens when sodium and fluorine "work together".

Go to the next page.

Compound Basics

1. Define molecule:
2. Define compound:

Ionic And Covalent Bonding Webquest Answer Key

**James M. Postma, Julian Roberts, J.
Leland Hollenberg**



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0.5 or 0.5 This polarizability parameter can be used to estimate the effective charge e_c defined by Callen which describes the splitting of longitudinal and transverse optic modes of vibration at $q = 0$. The result in GaAs is $e_c = 0.14$ or 0.04 in rough agreement with the experimental value of 0.20 or 0.03 . The experimental values for e_c may be used to estimate the gap parameters for each atom when complete data are available.

Author: Atoms, Electrons, Structure and Bonding Primrose Kitten, 2017-06-19

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Answers are provided for the sections where you need to work out the answers for yourself, not the sections where you are just filling things in from a video or website.

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