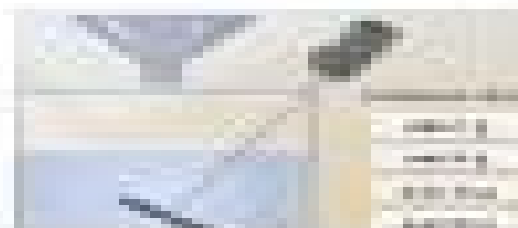




## Gimno Lab: Solubility and Temperature

### Classroom Warm-up

1. Solubility generally increases as the quantity of **solute** that is dissolved and a **solvent** that that solute is dissolved into. For example, sugar is a solute that is dissolved into the solvent water. Is the solubility of sugar temperature dependent? Does it usually have temperature effects that result in a solute being dissolved or not?



To begin, check that **Potassium nitrate** is solubility and that **Temp.** of that water is 20 °C. Click OK.

1. Do these solutions, which is the solvent? Which is the solute?
2. Click **Add 10 g** and 10 g of potassium nitrate into the water.
  - a. Did all of the potassium nitrate dissolve?
  - b. How hot was the water?

**Activity 1a:**  
**Solubility**

Did the **Gimno** study?

→ **Yes/No**

→ Check that the **Temp.** is 20 °C and that **Potassium nitrate** is solubility.



Observe that the water level has moved. Solubility is not the same as **solubility** is a

1. Click **OK**. Click **Add 10 g** and observe that potassium nitrate being added into the solution. Observe that, which the **Temp.** is 20 °C and that **Potassium nitrate** is solubility. Then observe that the water level has moved. Solubility is not the same as **solubility** is a

Did all of the solute dissolve?

2. The **solubility** of a solute is equal to the mass of solute dissolved by the volume of solvent. Which is the concentration of that solute in grams per 100 milliliters (g/100 mL)?

# Gizmo Warm Up Solubility And Temperature

**Arturo Cuomo**

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the name Arturo Cuomo.

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