



Name: Chelsey Williams Date: 3/31/22

Student Exploration: Stoichiometry

Vocabulary: Avogadro's number, balanced equation, cancel, coefficient, conversion factor, dimensional analysis, formula mass, molar mass, mole, molecular mass, stoichiometry

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

1. A 250 mL glass of orange juice contains 22 grams of sugar. How much sugar is in a two-liter (2,000 mL) bottle of orange juice? 176 grams
2. It requires two sticks of butter to make a batch of 20 cookies. How much butter will it take to make 150 cookies? 15 sticks of butter

Gizmo Warm-up

Just as a cook follows a recipe to decide how much of each ingredient to add, a chemist uses **stoichiometry** to determine the amounts of substances involved in chemical reactions. The Stoichiometry Gizmo allows you to try your hand at figuring out the amounts of reactants and products that take part in a chemical reaction.

To begin, check that this equation is shown:



1. Look at the **coefficients** (such as the "3" in 3CO) in front of each substance in the equation. The coefficients tell you how many molecules or atoms take part in a chemical reaction. In the spaces below, list the number of each molecule or atom in the equation:

Fe_2O_3 1

CO 3

Fe 2

CO_2 3

2. In a **balanced equation**, the same number of each kind of atom is shown on each side of the equation. Calculate the number of iron (Fe), oxygen (O), and carbon atoms (C).

New equation



New question

How many moles of carbon monoxide (CO) are required to react completely with 1.75 moles of iron (III) oxide (Fe_2O_3)?

Problems solved: 0



Gizmo Answers Stoichiometry

M Woodhall

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located on the right side of the slide, partially overlapping the light blue bar.

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Table of Contents Gizmo Answers Stoichiometry

1. Understanding the eBook Gizmo Answers Stoichiometry
 - The Rise of Digital Reading Gizmo Answers Stoichiometry
 - Advantages of eBooks Over Traditional Books
2. Identifying Gizmo Answers Stoichiometry
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Gizmo Answers Stoichiometry
 - User-Friendly Interface
4. Exploring eBook Recommendations from Gizmo Answers Stoichiometry
 - Personalized Recommendations
 - Gizmo Answers Stoichiometry User Reviews and Ratings
 - Gizmo Answers Stoichiometry and Bestseller Lists
5. Accessing Gizmo Answers Stoichiometry Free and Paid eBooks
 - Gizmo Answers Stoichiometry Public Domain eBooks
 - Gizmo Answers Stoichiometry eBook Subscription Services

- Gizmo Answers Stoichiometry Budget-Friendly Options
- 6. Navigating Gizmo Answers Stoichiometry eBook Formats
 - ePub, PDF, MOBI, and More
 - Gizmo Answers Stoichiometry Compatibility with Devices
 - Gizmo Answers Stoichiometry Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Gizmo Answers Stoichiometry
 - Highlighting and Note-Taking Gizmo Answers Stoichiometry
 - Interactive Elements Gizmo Answers Stoichiometry
- 8. Staying Engaged with Gizmo Answers Stoichiometry
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Gizmo Answers Stoichiometry
- 9. Balancing eBooks and Physical Books Gizmo Answers Stoichiometry
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Gizmo Answers Stoichiometry
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Gizmo Answers Stoichiometry
 - Setting Reading Goals Gizmo Answers Stoichiometry
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Gizmo Answers Stoichiometry
 - Fact-Checking eBook Content of Gizmo Answers Stoichiometry
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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