

# **FORMULAS WITH POLYATOMIC IONS**

Name \_\_\_\_\_

Matching the horizontal and vertical axes, write the formulas of the compounds with the following combination of ions. The first one is done for you.

	$\text{OH}^-$	$\text{NO}_3^-$	$\text{CO}_3^{2-}$	$\text{SO}_4^{2-}$	$\text{PO}_4^{3-}$
$\text{H}^+$	$\text{HOH}$ ( $\text{H}_2\text{O}$ )	$\text{HNO}_3$	$\text{H}_2\text{CO}_3$	$\text{H}_2\text{SO}_4$	$\text{H}_3\text{PO}_4$
$\text{Na}^+$					
$\text{Mg}^{+2}$					
$\text{NH}_4^+$					
$\text{Ca}^{+2}$					
$\text{K}^+$					
$\text{Al}^{+3}$					
$\text{Pb}^{+4}$					

# Formulas With Polyatomic Ions Worksheet

**SJ Ball**



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3 qqquad text distribute  $2 \cdot 3 \cdot 6x 10$  qqquad text apply the definition of a logarithm  $8 \cdot 6x 10$  qqquad text calculate  $2 \cdot 3 \cdot 18 \cdot 6x$  qqquad

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