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## NOVA: Hunting the Elements

## Part 1: Basic Chemistry

## Blank Periodic Table

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|                          |  |                     |  |                       |  |                     |  |                      |  |                       |  |                       |  |                     |  |                       |  |                      |  |                        |  |                        |  |                      |  |                      |  |                     |  |                      |  |                    |  |                    |  |                 |  |
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| 1<br>H<br>1.00794        |  |                     |  |                       |  |                     |  |                      |  |                       |  |                       |  |                     |  |                       |  | 2<br>He<br>4.002602  |  |                        |  |                        |  |                      |  |                      |  |                     |  |                      |  |                    |  |                    |  |                 |  |
| 3<br>Li<br>6.941         |  | 4<br>Be<br>9.012182 |  |                       |  |                     |  |                      |  |                       |  |                       |  |                     |  |                       |  |                      |  | 5<br>B<br>10.811       |  | 6<br>C<br>12.011       |  | 7<br>N<br>14.00643   |  | 8<br>O<br>15.999     |  | 9<br>F<br>18.998403 |  | 10<br>Ne<br>20.1797  |  |                    |  |                    |  |                 |  |
| 11<br>Na<br>22.98976928  |  | 12<br>Mg<br>24.304  |  |                       |  |                     |  |                      |  |                       |  |                       |  |                     |  |                       |  |                      |  | 13<br>Al<br>26.9815386 |  | 14<br>Si<br>28.08558   |  | 15<br>P<br>30.973762 |  | 16<br>S<br>32.06     |  | 17<br>Cl<br>35.453  |  | 18<br>Ar<br>39.948   |  |                    |  |                    |  |                 |  |
| 19<br>K<br>39.0983       |  | 20<br>Ca<br>40.078  |  | 21<br>Sc<br>44.955912 |  | 22<br>Ti<br>47.88   |  | 23<br>V<br>50.9415   |  | 24<br>Cr<br>51.9961   |  | 25<br>Mn<br>54.938045 |  | 26<br>Fe<br>55.845  |  | 27<br>Co<br>58.933195 |  | 28<br>Ni<br>58.6934  |  | 29<br>Cu<br>63.546     |  | 30<br>Zn<br>65.38      |  | 31<br>Ga<br>69.723   |  | 32<br>Ge<br>72.64    |  | 33<br>As<br>74.9216 |  | 34<br>Se<br>78.96    |  | 35<br>Br<br>79.904 |  | 36<br>Kr<br>83.798 |  |                 |  |
| 37<br>Rb<br>85.4678      |  | 38<br>Sr<br>87.62   |  | 39<br>Y<br>88.90584   |  | 40<br>Zr<br>91.224  |  | 41<br>Nb<br>92.90638 |  | 42<br>Mo<br>95.94     |  | 43<br>Tc<br>98.906251 |  | 44<br>Ru<br>101.07  |  | 45<br>Rh<br>102.9055  |  | 46<br>Pd<br>106.3631 |  | 47<br>Ag<br>107.8642   |  | 48<br>Cd<br>112.411    |  | 49<br>In<br>114.818  |  | 50<br>Sn<br>118.710  |  | 51<br>Sb<br>121.757 |  | 52<br>Te<br>127.6    |  | 53<br>I<br>126.905 |  | 54<br>Xe<br>131.29 |  |                 |  |
| 55<br>Cs<br>132.90545196 |  | 56<br>Ba<br>137.327 |  | 57-70<br>*            |  | 71<br>Lu<br>174.967 |  | 72<br>Hf<br>178.49   |  | 73<br>Ta<br>180.94788 |  | 74<br>W<br>183.84     |  | 75<br>Re<br>186.207 |  | 76<br>Os<br>190.23    |  | 77<br>Ir<br>192.222  |  | 78<br>Pt<br>195.084    |  | 79<br>Au<br>196.966569 |  | 80<br>Hg<br>200.59   |  | 81<br>Tl<br>204.3833 |  | 82<br>Pb<br>207.2   |  | 83<br>Bi<br>208.9804 |  | 84<br>Po<br>209    |  | 85<br>At<br>210    |  | 86<br>Rn<br>222 |  |
| 87<br>Fr<br>223          |  | 88<br>Ra<br>226     |  | 89-102<br>(*) (*)     |  | 103<br>Lr<br>260    |  | 104<br>Rf<br>261     |  | 105<br>Db<br>262      |  | 106<br>Sg<br>266      |  | 107<br>Bh<br>264    |  | 108<br>Hs<br>277      |  | 109<br>Mt<br>268     |  | 110<br>Uun<br>288      |  | 111<br>Uuu<br>289      |  | 112<br>Uub<br>285    |  | 114<br>Uuq<br>289    |  |                     |  |                      |  |                    |  |                    |  |                 |  |

<sup>a</sup> Lanthanide series.

|                 |                 |                 |                 |                 |                 |                 |                 |                 |                 |                 |                  |                  |                  |
|-----------------|-----------------|-----------------|-----------------|-----------------|-----------------|-----------------|-----------------|-----------------|-----------------|-----------------|------------------|------------------|------------------|
| 57<br><b>La</b> | 58<br><b>Ce</b> | 59<br><b>Pr</b> | 60<br><b>Nd</b> | 61<br><b>Pm</b> | 62<br><b>Sm</b> | 63<br><b>Eu</b> | 64<br><b>Gd</b> | 65<br><b>Tb</b> | 66<br><b>Dy</b> | 67<br><b>Ho</b> | 68<br><b>Er</b>  | 69<br><b>Tm</b>  | 70<br><b>Yb</b>  |
| 89<br><b>Ac</b> | 90<br><b>Th</b> | 91<br><b>Pa</b> | 92<br><b>U</b>  | 93<br><b>Np</b> | 94<br><b>Pu</b> | 95<br><b>Am</b> | 96<br><b>Cm</b> | 97<br><b>Bk</b> | 98<br><b>Cf</b> | 99<br><b>Es</b> | 100<br><b>Fm</b> | 101<br><b>Md</b> | 102<br><b>No</b> |

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**Gold - Au**

1. Write the number of subatomic particles in gold:
  - a. Protons:
  - b. Neutrons:
  - c. Electrons:
2. Give one property of gold.
3. How much gold is in one ton of the mined rock?

# Hunting The Elements Answers

**Clemens Wendtner**



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