
Chemistry

Checklist & Study Guide

Lindbergh High School
2014-2015
K. Smith, Room 223

Overview

The following topics are reviewed from the physical science course. Lectures and in-class activities are designed so that students focus more on the application of the concepts and big ideas. For the first exam students are expected to demonstrate their knowledge in the following areas:

Big Ideas

The semester final assesses your ability to demonstrate your learning on the following big ideas:

1. Math & Measurement: **All measurements have a degree of uncertainty.**
2. Matter- Classification, Properties, and Changes: **Physical material in our universe is made of matter.**
3. The Structure of the Atom: **Atoms are the fundamental building blocks of matter.**

Learning Objectives & Outcomes

Math & Measurement	I CAN:	Assignments
All measurements have a degree of uncertainty – Measurements are not exact numbers	Determine the number of significant figures in a measurement	Significant Figures Practice
	Use the correct number of significant figures in reporting measurements and the results of calculations.	Pg. 95: # 56, 80, 63, 59, 74
	Use the appropriate SI units for length, mass, volume	Scientific Notation Practice I
	Know the following SI prefixes: K H D (Base) d c m μ , n	Extra Practice: Pg. 99 # 1, 2
	Analyze the accuracy and precision of data	

Questions?

High School Chemistry Study Guide

D Keegan



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Table of Contents High School Chemistry Study Guide

1. Understanding the eBook High School Chemistry Study Guide
 - The Rise of Digital Reading High School Chemistry Study Guide
 - Advantages of eBooks Over Traditional Books
2. Identifying High School Chemistry Study Guide
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an High School Chemistry Study Guide
 - User-Friendly Interface
4. Exploring eBook Recommendations from High School Chemistry Study Guide
 - Personalized Recommendations
 - High School Chemistry Study Guide User Reviews and Ratings
 - High School Chemistry Study Guide and Bestseller Lists
5. Accessing High School Chemistry Study Guide Free and Paid eBooks
 - High School Chemistry Study Guide Public Domain eBooks
 - High School Chemistry Study Guide eBook Subscription Services
 - High School Chemistry Study Guide Budget-Friendly Options
6. Navigating High School Chemistry Study Guide eBook Formats

- ePub, PDF, MOBI, and More
- High School Chemistry Study Guide Compatibility with Devices
- High School Chemistry Study Guide Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of High School Chemistry Study Guide
 - Highlighting and Note-Taking High School Chemistry Study Guide
 - Interactive Elements High School Chemistry Study Guide
- 8. Staying Engaged with High School Chemistry Study Guide
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers High School Chemistry Study Guide
- 9. Balancing eBooks and Physical Books High School Chemistry Study Guide
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection High School Chemistry Study Guide
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine High School Chemistry Study Guide
 - Setting Reading Goals High School Chemistry Study Guide
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of High School Chemistry Study Guide
 - Fact-Checking eBook Content of High School Chemistry Study Guide
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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