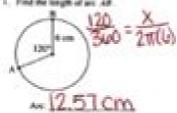
Arc Length-Sector Area Name: ((U



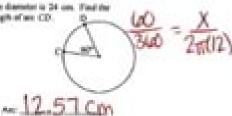
Wells Workshoot (W2)

1. Find the length of my Alf.



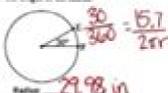
2. The diameter is 24 cm. Faul than length of our CD.

Danc

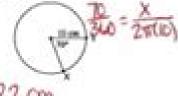


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3. The bright of arc EF is Ellis Find the bright of the rights.



4. Find the length of art. XY



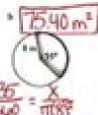
1. A circle has an art whose exposure in \$67 and whose bought in \$25/\$ other in the dissense. of the similar

6. A circle has a cocumilierare whose length in 75%. Find the length of an art whose comed angle is 907.

1. Find the measure of the control angle of an arc if to length in 1808 and the radius is 18.

Arc Length-Sector Area







8. The was of a circle is 225e square inches. Find the area of the proton whose control segle in 40°

16. The control angle of a sector is 60° and the area of the circle in 144x. What is the area of the index?

Find the area of the sector whose certain engle in 128'

12. Find the radius of a credit which has a notice area of 250-hour central angle in W.

$$\frac{90}{360} = \frac{28.3}{\pi r^2}$$
 $r^2 = 36.03$

77' and the sector has an own of 102. Find the redice.

a stone of its area in the and the radius to it.

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web feb 13 2023 we will revisit this today and go on to see that if we use a special kind of thermochemical reaction called the standard enthalpy of formation we can calculate enthalpies of reactions without having to manipulate a series of individual thermochemical equations for each step

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web consider the thermochemical equation below 4 no g 6 h 2 o l 4 nh 3 g 5 o 2 g rh 1170 kj a is this reaction endothermic or exothermic endothermic rh gt 0 b how much heat will be absorbed or released if 10 g of no is reacted with 10 g h 2 o $^{\circ}$

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web sep 5 2023 3 carry out a thermochemical calculation on all these species at their optimized geometries 4 sum up the enthalpies free energy of the products and the reagents keeping into account the

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web answer a 1 explanation the entire reaction was multiplied by 2 therefore Δh must also be halved 3 answer c explanation option c is incorrect because 566 kj mol is the Δh per mole of o2 and not for 2 moles even though options b and d are expressed in grams they are still correct because 56

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web q 140 g 4 184 j 16oc g oc q 9372 16 j 35 5 g x 1 mole kcn 0 5453 moles 65 1 g 0 5453 mole moles 5 when a 15 g sample of srcl2 dissolves in 80 g of water in a calorimeter the temperature drops from 42 oc to 15 c calculate the Δh for the process thermochemistry answers and explanations integrated mcat - Feb 11 2023

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web jan 10 2021 Δh 27 4 kj mol and Δs 29 8 j k sulfur trioxide so 3 is produced in large quantities in the industrial synthesis of sulfuric acid sulfur dioxide is converted to sulfur trioxide by reaction with oxygen gas write a balanced chemical equation for the reaction of so 2 with o 2 g and determine its Δg