

Cellular Respiration

SHORT ANSWER

Formulation of Different Sugars by Yeast Cells

Lab Results:

- The graph of the sugar fermented by yeast. All listed other factors for determine the amount of CO₂ produced.

Sugars Type	Substrate Amount and structure (mol)	Product Yielded at 15 minutes (mol)	Amount of CO ₂ produced at that condition
glucose	0.001	0.01	0.01
fructose	0.001	0.01	0.01
sucrose	0.001	0.01	0.01
maltose	0.001	0.01	0.01

- The graph of the sugar fermented by yeast. All listed other factors for determine the rate of sugar fermented per volume of starting fermentation. They should have 1 for substrate for glucose from the multiple choice word bank.
 - To calculate the value of CO₂ produced per (for substrate) available, assuming P = 1 atm.
 - at 15 min
 - value P is the atmosphere pressure at the lab. It is the volume in liter of CO₂ mole produced at volume of CO₂. If the time given converted to CO₂ is, atmosphere liter, and P is the temperature in Kelvin.
 - To calculate the value of sugar converted use the following equation:

$$n = \frac{PV}{RT}$$
 - value n is the amount of moles of sugar converted, or in this condition of moles of CO₂ produced, and it is the amount of sugar in that sugar.
 - To calculate the rate of sugar fermented per volume use the following equation:

$$r = \frac{2000 \cdot \text{mol} \cdot \text{g}}{t}$$
 - value r is the rate of sugar fermented per volume, or is the amount of moles of sugar converted, or is the mole fraction weight of the sugar in grams, which is the sugar in the fermentation.

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