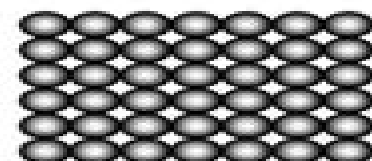


Everything is made up of particles that are too small to see. The three states of matter are **SOLID**, **LIQUID** and **GAS**. They all have different properties due to the arrangement and movement of their particles.

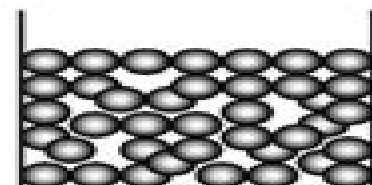
Solids.

The particles are held tightly together by strong forces. They make small vibrations but they stay in place. This gives solids a definite shape and volume. Solids are **DENSE** (heavy) and they can not be compressed (squashed) easily because the particles are already packed closely together.



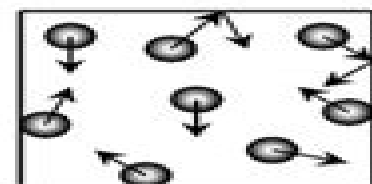
Liquids.

A liquid can flow because the particles can move past each other. The particles are still held closely together by strong forces. Liquids are **DENSE** and they can not be compressed easily. A liquid can change its shape but not its volume.



Gases.

There are only very weak forces between the particles which are far apart. The particles move around very quickly and bounce off each other. Gases have a low density (they are very light) and they do not have a definite shape or volume.



Exercise - Complete the spaces in the table below.

Property	Solids	Liquids	Gases
Density (heavy or light)	High density (heavy)		Low density (light)
How easy are they to compress (squash)?	Hard		Easy
Do they flow?		Yes	Yes
Do they keep the same shape?		No	
Do they keep the same volume?	Yes		

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Table of Contents Ks3 Science Revision Worksheets Special Edition

1. Understanding the eBook Ks3 Science Revision Worksheets Special Edition
 - The Rise of Digital Reading Ks3 Science Revision Worksheets Special Edition
 - Advantages of eBooks Over Traditional Books
2. Identifying Ks3 Science Revision Worksheets Special Edition
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Ks3 Science Revision Worksheets Special Edition
 - User-Friendly Interface
4. Exploring eBook Recommendations from Ks3 Science Revision Worksheets Special Edition
 - Personalized Recommendations
 - Ks3 Science Revision Worksheets Special Edition User Reviews and Ratings
 - Ks3 Science Revision Worksheets Special Edition and Bestseller Lists
5. Accessing Ks3 Science Revision Worksheets Special Edition Free and Paid eBooks
 - Ks3 Science Revision Worksheets Special Edition Public Domain eBooks
 - Ks3 Science Revision Worksheets Special Edition eBook Subscription Services

- Ks3 Science Revision Worksheets Special Edition Budget-Friendly Options
- 6. Navigating Ks3 Science Revision Worksheets Special Edition eBook Formats
 - ePub, PDF, MOBI, and More
 - Ks3 Science Revision Worksheets Special Edition Compatibility with Devices
 - Ks3 Science Revision Worksheets Special Edition Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Ks3 Science Revision Worksheets Special Edition
 - Highlighting and Note-Taking Ks3 Science Revision Worksheets Special Edition
 - Interactive Elements Ks3 Science Revision Worksheets Special Edition
- 8. Staying Engaged with Ks3 Science Revision Worksheets Special Edition
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Ks3 Science Revision Worksheets Special Edition
- 9. Balancing eBooks and Physical Books Ks3 Science Revision Worksheets Special Edition
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Ks3 Science Revision Worksheets Special Edition
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Ks3 Science Revision Worksheets Special Edition
 - Setting Reading Goals Ks3 Science Revision Worksheets Special Edition
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Ks3 Science Revision Worksheets Special Edition
 - Fact-Checking eBook Content of Ks3 Science Revision Worksheets Special Edition
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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