



Hunter 30 2 Sailboat Manual

Guillaume Favre



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web when the atoms in a bond are not the same the bonding electrons are shared and the bond is electronegativities equally
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atoms but in most cases those atoms aren t just floating around individually

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molecules and ions contain polar bonds which of these molecules and ions have dipole moments a clf 5 b clo 2 clo 2 c tecl2 4
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web the relative electronegativities of the bonded atoms is h c cl and so the bond moments all point toward the cl end of the
molecule and sum to yield a considerable dipole moment the molecules are relatively polar for molecules of high symmetry
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