

Chemistry – Unit 7 Review

Chemical Reaction Model

1. Describe key characteristics of all chemical reactions, including the role of energy. Explain how a balanced equation represents these features (include an example).

In chemical reactions, atoms of the reactants recombine to form new substances in the products. There is almost always some transfer between the thermal and chemical energy accounts as well as an exchange of thermal energy via heating (Q) between the system and surroundings. The balanced chemical equation represents this process symbolically, showing that the atoms in the products are the same as those in the reactants



Note: there are 4 atoms of H and 2 atoms of O on each side of the equation.

Applying the Model

2. Balance the reaction below by drawing any additional reactant and/or product molecules that would be needed.



Write the **balanced equation** for this reaction.



Which of these molecules would be called reactants?

DB₂, CD₂, and BAC are reactants

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Charles R. Hadlock



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Vishwanath, 2006 Contents Introduction Theoretical Framework Review of Research Literature Models of Teaching in Environmental Education EE Methodology Data Analysis and Interpretation Summary and Conclusions

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Supplementary Material and Solutions Manual for Mathematical Modeling in the Environment Charles R. Hadlock,2020-05-05 This manual is meant to provide supplementary material and solutions to the exercises used in Charles Hadlock s textbook Mathematical Modeling in the Environment The manual is invaluable to users of the textbook as it contains complete solutions and often further discussion of essentially every exercise the author presents in his book This includes both the mathematical computational exercises as well as the research questions and investigations Since the exercises in the textbook are very rich in content rather than simple mechanical problems and cover a wide range most readers will not have the time to work out every one on their own Readers can thus still benefit greatly from perusing solutions to problems they have at least thought about briefly Students using this manual still need to work out solutions to research questions using their own sources and adapting them to their own geographic locations or to numerical problems using their own computational schemes so this manual will be a useful guide to students in many course contexts Enrichment material is included on the topics of some of the exercises Advice for teachers who lack previous environmental experience but who want to teach this material is also provided and makes it practical for such persons to offer a course based on these volumes This book is the essential companion to Mathematical Modeling in the Environment

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