Chemistry Unit 8 Worksheet 4 Samples of Every Kind of Problem

On a separate sheet of paper, write a complete solution to each of the problems below. Follow the procedure outlined in class. Be sure to circle your final answer.

 Calculate the number of moles of potassium chlorate, KClO₃ (s), that must decompose to produce potassium chloride, KCl (s), and 1.8 moles of oxygen gas.

Equation:	2 KClO ₃ →	2 KCI +	3 O2
Before:	1.2	0	0
Change:	-1.2	+1.2	+1.8
After:	0	1.2	1.8

1.2 moles KClO₃ needed

2. In a single displacement reaction, magnesium metal reacts with hydrochloric acid to produce magnesium chloride and hydrogen gas. How many moles of hydrochloric acid are needed to completely react with 2.43 g of magnesium?

Equation:	Mig +	2 HCl ->	MgCl ₂ +	H ₂
Before:	0.100	- 7	0	0
Change:	-0.100	-0.200 +	0.160 +	0.100
Afters	0		0.100	0.100

0,200 moles HCI needed

0.73 g x 1 mole = .0040 mole 183 g

3. Ethane, C₂H₆ reacts with oxygen gas to produce carbon dioxide gas and water vapor. What mass of oxygen gas is required to react with 2.20 moles of ethane?

Equation:	2 C2H6	+ 7 O ₂	→ 4 CO₂	+ 6 H ₂ O
Before:	2.20	36.56	0	0
Change:	-2.20	- 7.70	+ 4,40	+ 6.60
After:	0	X 5	4.40	6.60

O₂ needed: 7.70 mole x 32.0 g = 246 g O₂ required 1 mole

 Determine the mass of sodium nitrate produced when 0.73 g of nickel (II) nitrate reacts with sodium hydroxide according to the following equation: Ni(NO3)2 + 2 NaOH --> Ni(OH)2 + 2 NaNO3

 Equation:
 Ni(NO₃)₂ + 2 NaOH
 → Ni(OH)₂ + 2 NaNO₃

 Before:
 0.0040
 XS
 0
 0

 Change:
 -0.0040
 -0.0080
 +0.0040
 +0.0080

 After:
 0
 XS
 0.0040
 0.0080

NaNO, produced: 0.0080 mole x 85.0 g = 0.68 g 1 mole

Modeling Chemistry Unit 8

Syed Ahmad Imtiaz

Modeling Chemistry Unit 8:

Molecular Modelling Peter Bladon, John E. Gorton, Robert B. Hammond, 2012 This book is a practical easy to use guide for readers with limited experience of molecular modelling Unlike many other textbooks in this field the authors avoid extensive discussion around complex mathematical foundations behind the methods choosing instead to provide the reader with the choice of methods themselves Modeling of Combustion Systems Joseph Colannino, 2025-02-20 Modeling of Combustion Systems provides accessible instruction for the operator or engineer to model combustion responses such as NOx CO flame dimensions or heat flux as a function of input factors such as excess oxygen firing rate furnace temperature and fuel type Retaining the emphasis on practical semiempirical modeling of boilers process heaters and reactors such as ethylene cracking units and ammonia and hydrogen reformers the second edition expands the approach to include combustion air fans and representation of complex models direct numerical simulations or computational fluid dynamics in concise forms It also shows how to design experiments to collect the most data with the fewest experiments and model real unit behavior with engineering accuracy This book is intended for technicians operators and engineers working with fired equipment such as boilers reactors and process heaters It includes 65 fully worked examples in MS Excel with detailed instructions All example spreadsheets are available for download from the publisher's website **Community (ChemCom)** American Chemical Society, 2011-06-17 Touted as the most successful NSF funded project published Chemistry in the Community ChemCom by the American Chemical Society ACS offers a meaningful and memorable chemistry program for all levels of high school students ChemCom covers traditional chemistry topics within the context of societal issues and real world scenarios Centered on decision making activities where students are responsible for generating data in an investigating analyzing that data and then applying their chemistry knowledge to solve the presented problem The text is intensively laboratory based with all 39 of the investigations integrated within the text not separate from the reading With the ChemCom program students learn more organic and biochemistry more environmental and industrial chemistry and more on the particulate nature of matter than other textbooks all within the relevance of solving problems that arise in everyday life Meticulously updated to meet the needs of today s teachers and students the new sixth edition of ChemCom adheres to the new science framework as well as the forthcoming next generation of science standards Incorporating advances in learning and cognitive sciences ChemCom's wide ranging coverage builds upon the concepts and principles found in the National Science Education Standards Correlations are available showing how closely aligned ChemCom is to these and other state standards **Comprehensive Inorganic Chemistry II**, 2013-07-23 Comprehensive Inorganic Chemistry II Nine Volume Set reviews and examines topics of relevance to today s inorganic chemists Covering more interdisciplinary and high impact areas Comprehensive Inorganic Chemistry II includes biological inorganic chemistry solid state chemistry materials chemistry and nanoscience The work is designed to follow on with a different viewpoint and

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