

Chemistry – Unit 7 Review

Chemical Reaction Model

1. Describe key characteristics of all chemical reactions, including the role of energy. Explain how a balanced equation represents these features (include an example).

In chemical reactions, atoms of the reactants recombine to form new substances in the products. There is almost always some transfer between the thermal and chemical energy accounts as well as an exchange of thermal energy via heating (Q) between the system and surroundings. The balanced chemical equation represents this process symbolically, showing that the atoms in the products are the same as those in the reactants



Note: there are 4 atoms of H and 2 atoms of O on each side of the equation.

Applying the Model

2. Balance the reaction below by drawing any additional reactant and/or product molecules that would be needed.



Write the **balanced equation** for this reaction.



Which of these molecules would be called reactants?

DB₂, CD₂, and BAC are reactants

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Ensheng Dong



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1. Understanding the eBook Modeling Chemistry U3 Review V2 2 Answers
 - The Rise of Digital Reading Modeling Chemistry U3 Review V2 2 Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Modeling Chemistry U3 Review V2 2 Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Modeling Chemistry U3 Review V2 2 Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Modeling Chemistry U3 Review V2 2 Answers

- Personalized Recommendations
 - Modeling Chemistry U3 Review V2 2 Answers User Reviews and Ratings
 - Modeling Chemistry U3 Review V2 2 Answers and Bestseller Lists
5. Accessing Modeling Chemistry U3 Review V2 2 Answers Free and Paid eBooks
 - Modeling Chemistry U3 Review V2 2 Answers Public Domain eBooks
 - Modeling Chemistry U3 Review V2 2 Answers eBook Subscription Services
 - Modeling Chemistry U3 Review V2 2 Answers Budget-Friendly Options
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 - Modeling Chemistry U3 Review V2 2 Answers Enhanced eBook Features
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 - Adjustable Fonts and Text Sizes of Modeling Chemistry U3 Review V2 2 Answers
 - Highlighting and Note-Taking Modeling Chemistry U3 Review V2 2 Answers
 - Interactive Elements Modeling Chemistry U3 Review V2 2 Answers
 8. Staying Engaged with Modeling Chemistry U3 Review V2 2 Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Modeling Chemistry U3 Review V2 2 Answers
 9. Balancing eBooks and Physical Books Modeling Chemistry U3 Review V2 2 Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Modeling Chemistry U3 Review V2 2 Answers
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Modeling Chemistry U3 Review V2 2 Answers
 - Setting Reading Goals Modeling Chemistry U3 Review V2 2 Answers
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Modeling Chemistry U3 Review V2 2 Answers

- Fact-Checking eBook Content of Modeling Chemistry U3 Review V2 2 Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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