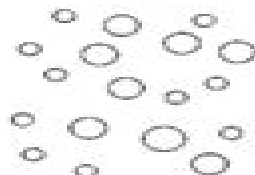
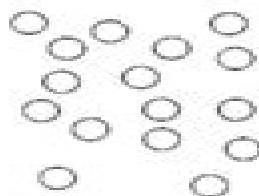


Chapter 2 Matter Study Guide

Answer the following as completely as possible. Use your notebook assignments to help you prepare for your second test. If you cannot find the information in your notebook, then refer to your textbook.

1. Describe a heterogeneous mixture and give an example. A mixture where you can see different parts example M&Ms
2. Describe a homogeneous mixture and give an example. A mixture where the pieces appear to look similar throughout because substances are so evenly mixed. An example is Salt water
3. What makes up elements? A pure substance that cannot be broken down into any other substance
4. What makes up compounds? a substance made of two or more elements
5. What is a mixture? Made up of more than one kind of matter Can be separated physically by: Evaporation, Filtering, Sorting, Electric Charge
6. Label the pictures as compound, mixture or element. Each circle represents an atom. If two atoms are touching then they are bonded together. #1 Element, #2 mixture, #3 compound



7. What is a Physical Property? Properties that do not change the chemical nature of matter
8. What is a Chemical Property? Properties that describe the way a substance may change or react to form another substance
9. Give Examples of:

Physical Properties	Chemical Properties
Color-Odor-Luster- Malleability- Conductivity- Hardness- Melting Point/Freezing Point- Boiling Point- Density- Volume-Length-	Combustibility-reactivity

10. Give examples of phase changes. Melting, boiling, evaporation, condensation,
11. What is Matter? Matter is anything that has mass and volume, It occupies space, composed of atoms
12. How can matter be broken down into smaller groups? *A chart is fine for this one.*

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