

EASEL BALANCING & REACTION QUIZ

Multiple Choice: Select the most correct answer. (15 R)

1. State the type of reaction: $3 \text{AgNO}_3 + \text{K}_3\text{PO}_4 \rightarrow \text{Ag}_3\text{PO}_4 + 3 \text{KNO}_3$

- A) synthesis B) decomposition C) single displacement
D) double displacement E) combustion F) no reaction

2. State the type of reaction: $\text{Ca} + 2 \text{HCl} \rightarrow \text{CaCl}_2 + \text{H}_2$

- A) synthesis B) decomposition C) single displacement
D) double displacement E) combustion F) no reaction

3. State the type of reaction: $2 \text{NaClO}_3 \rightarrow 2 \text{NaCl} + 3 \text{O}_2$

- A) synthesis B) decomposition C) single displacement
D) double displacement E) combustion F) no reaction

4. State the type of reaction: $\text{P}_4 + 5 \text{O}_2 \rightarrow 2 \text{P}_2\text{O}_5$

- A) synthesis B) decomposition C) single displacement
D) double displacement E) combustion F) no reaction

5. Which of the following type(s) of chemical reactions produces water and an ionic compound?

- A) single displacement B) neutralization C) decomposition D) synthesis

6. Chemical equations must be balanced to satisfy _____.

- A) the law of definite proportions C) the law of conservation of mass
B) the law of multiple proportions D) Avogadro's principle

7. To balance a chemical equation, it may be necessary to adjust the:

- A) coefficient(s) B) subscript(s) C) formulae of the product(s) D) number of products

8. Propane gas, C_3H_8 , burns in excess oxygen gas. When the equation for this reaction is correctly balanced and all coefficients are reduced to their lowest whole-number terms, the coefficient for CO_2 is:

- A) 1 B) 2 C) 3 D) 4 E) 5 F) 6

9. Which equation is not balanced?

- A) $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$ B) $4\text{H}_2 + 3\text{O}_2 \rightarrow 4\text{H}_2\text{O}$
C) $\text{H}_2 + \text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O} + \text{H}_2\text{O}$ D) $2\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O}$

10. $\text{C}_6\text{H}_8\text{O}_4\text{S}_2 + \text{O}_{20} \rightarrow \text{CO}_2 + \text{SO}_2 + \text{H}_2\text{O}$

When the equation above is balanced and all coefficients are reduced to their lowest whole-number terms, the coefficient for $\text{C}_6\text{H}_8\text{O}_4\text{S}_2$ is:

- A) 1 B) 2 C) 6 D) 10 E) 12

Mixed Type Practice For Predicting Reactions Answers

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Organic Chemistry T. W. Graham Solomons, Craig B. Fryhle, Scott A. Snyder, 2022-01-19 Organic Chemistry 13th Edition continues Solomons Fryle and Snyder s tradition of excellence in teaching and preparing students for success in both the classroom and beyond Central to the authors is their approach in emphasizing organic chemistry s relationship between structure and reactivity To accomplish

this the content is organized in a way that combines the most useful features of a functional group approach with one largely based on reaction mechanisms The authors philosophy is to emphasize mechanisms and their common aspects as often as possible and at the same time use the unifying features of functional groups as the basis for most chapters The structural aspects of the authors approach show students what organic chemistry is Mechanistic aspects of their approach show students how it works And wherever an opportunity arises the authors show students what it does in living systems and the physical world around us

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