

## Equilibrium HOMEWORK

100

5. Write the equilibrium constant expressions for the following reactions. Use pressures for gases and concentrations for solutions.



**\*\*\*Caution\*\*\*: The 1978-1994 and 1995-2010  
NIAID data are not included in the equilibrium  
estimation.**

2. The fraction before taken place in a 1-liter container. It produced a mixture that contains 0.0001 (0.01%), 0.0001 (0.01%), and 0.0001 (0.01%) of the substance in an amount of 1.000 g.



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2010-2011

**Figure 1**

8. Based on the equilibrium constant, do these two reactions favor reactants or products? Answer: at 1,000 K?

repeated. However, because  $\beta$  is fixed, this may not be sufficient to ensure that products are equivalent.

15. The reaction below takes place in a 1-liter container. It produces a gaseous mixture that contains 0.20 mole  $\text{CO}_2$ , 0.80 mole  $\text{H}_2$ , 0.20 mole  $\text{H}_2\text{O}$ , and 0.50 mole  $\text{CO}$ . This mixture is at equilibrium at 1000 K.



18. <http://www.fda.gov/cder/rdmt/rdmt.htm> accessed Nov 2006

\*Don't forget to send us your name, address, phone number, and e-mail address.

1000

doi:10.1017/S0022292412001771

b. Based on the equilibrium constant, is this reaction favored forward or reverse? Answered at 1,000 K?

Statistical significance: Variables in  $\Sigma$  are less than one, indicating they are more uncertain. Also, gradient, or significance

4. The reaction below takes place in a 50.0-Liter container. It produces a gaseous mixture that contains 2.00mol  $H_2$ , 8.00mol  $H_2O$ , and 1.00mol  $NH_3$ . This mixture is at equilibrium at 1.00atm.



10. <https://doi.org/10.1002/for>

<sup>a</sup> Because the polystyrenes grafted on the *SACCLITE* and *LTITE*, grafted are represented as *SACCLITE*-*g* and *LTITE*-*g*.

10-10-2014

\*Indicates that 1987-92 cannot be entered into an exponential regression. It must be 1987-92, which is easily checked by hand.

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16. Based on the equilibrium constant, is this reaction reactant-favored or product-favored at 1200 K?

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# Mixed Review Chemical Equilibrium Answers

**Ralph Thomas Myers**



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Principles of Pollution Abatement S. Jørgensen, 2000-09-21 This is a much revised and expanded version of the author s previous book Principles of Environmental Science 1988 also published by Elsevier It incorporates new tools such as ecotechnology cleaner technology life cycle analysis and environmental risk assessment which have changed environmental management strategy radically over the last decade Sven J rgensen has worked at the Royal

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