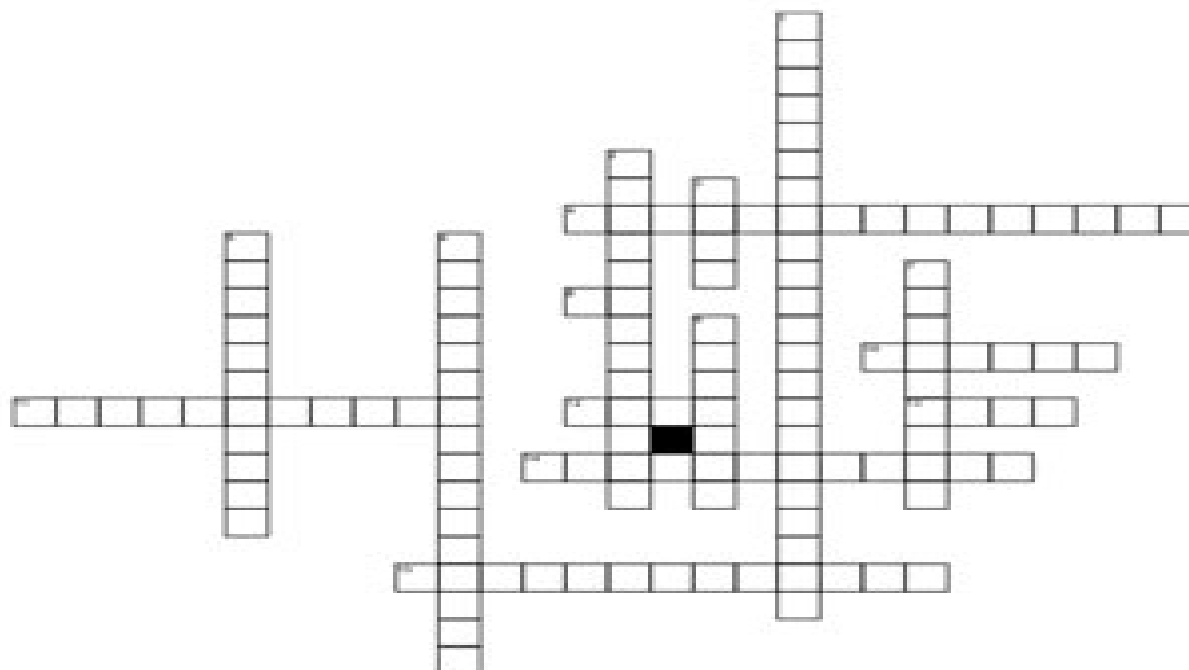


Chapter 15 Chemical Compounds



Across

4. A covalently bonded compound that contains carbon
 8. A value that is used to express the acidity or basicity (alkalinity) of a system
 10. a type of biochemical that does not dissolve in water; fats and steroids
 11. An organic compound composed only of carbon and hydrogen
 12. An ionic compound that forms when a metal atom replaces the hydrogen of an acid
 13. Any compound that increases the number of hydronium ions when dissolved in water

14. Combining of atoms to form molecules or compounds

15. A compound made of oppositely charged ions

Down

1. The reaction of an acid and a base to form a neutral solution of water and a salt
 2. A class of energy-giving nutrients that includes sugars, starches, and fiber; composed of one or more simple sugars bonded together
 3. Any compound that increases the number of hydroxide ions when dissolved in water

5. A molecule made up of subunits called nucleotides

6. A chemical compound formed by the sharing of electrons

7. A compound that can reversibly change color depending on the pH of the solution or other chemical change

9. A molecule that is made up of amino acids and that is needed to build and repair body structures and to regulate processes in the body

Word Bank

Chemical bond

Lipids

pH

Base

Ionic compound

Protein

Salt

Indicator

Covalent compound

Nucleic acid

Organic compound

Acid

Carbohydrates

Neutralization reaction

Hydrocarbon

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the speed of p is less than the speed of q the sum of the final kinetic energies of p and q is zero the sum of the final momentum of p and q is zero 2 1 5 the diagram below shows the electric field pattern due to two point charges x and y which one of the following represents the charge on x and y respectively point charge x point charge y

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november 2014 physical sciences paper i time 3 hours 200 marks please read the following instructions carefully 1 this paper consists of a question paper of 16 pages a yellow answer booklet of 4 pages i iv and a green data and formulae sheet of 2 pages i ii please make sure that your question paper is complete