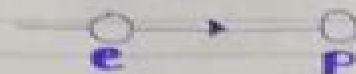


a) **Force due to electron:** As electron is negatively charged and proton is positively charged so, the direction of force on central proton due to electron is "**Leftward**".

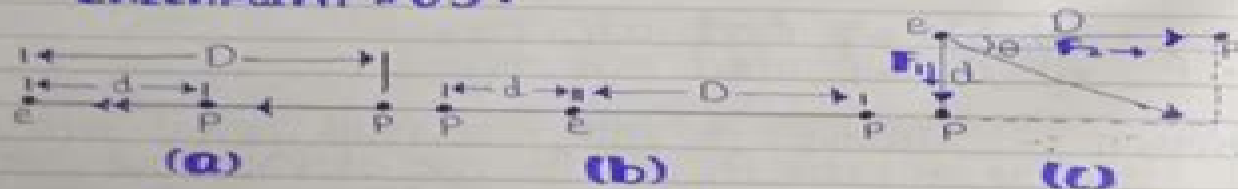


b) **Force due to proton:** As both have same charges so, they will repel each other. The direction of force on central proton due to other proton is "**Leftward**".



c) **Net Force:** The direction of net force on central proton will be "**Leftward**" as the force due to electron as well as other proton on central proton is in leftward direction.

CHECKPOINT # 03 :



⇒ a) **Ranking Arrangements:** According to the magnitude of the net electrostatic forces on the electron due to protons, the arrangements will be as: **Figure (a)** is first as it has largest net electrostatic force as both protons exert attractive force on electron in same direction.

⇒ b) **Angle:** Figure (c) shows us that $F_1 > F_2$ and

$$\tan \theta = \frac{F_1}{F_2} \Rightarrow \theta = \tan^{-1} \frac{F_1}{F_2}$$

We can conclude $\theta > 45^\circ$

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Table of Contents Halliday Resnick Check Point

1. Understanding the eBook Halliday Resnick Check Point
 - The Rise of Digital Reading Halliday Resnick Check Point
 - Advantages of eBooks Over Traditional Books
2. Identifying Halliday Resnick Check Point
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Halliday Resnick Check Point
 - User-Friendly Interface
4. Exploring eBook Recommendations from Halliday Resnick Check Point
 - Personalized Recommendations
 - Halliday Resnick Check Point User Reviews and Ratings
 - Halliday Resnick Check Point and Bestseller Lists
5. Accessing Halliday Resnick Check Point Free and Paid eBooks
 - Halliday Resnick Check Point Public Domain eBooks
 - Halliday Resnick Check Point eBook Subscription Services
 - Halliday Resnick Check Point Budget-Friendly Options
6. Navigating Halliday Resnick Check Point eBook Formats

- ePub, PDF, MOBI, and More
- Halliday Resnick Check Point Compatibility with Devices
- Halliday Resnick Check Point Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Halliday Resnick Check Point
 - Highlighting and Note-Taking Halliday Resnick Check Point
 - Interactive Elements Halliday Resnick Check Point
- 8. Staying Engaged with Halliday Resnick Check Point
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Halliday Resnick Check Point
- 9. Balancing eBooks and Physical Books Halliday Resnick Check Point
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Halliday Resnick Check Point
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Halliday Resnick Check Point
 - Setting Reading Goals Halliday Resnick Check Point
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Halliday Resnick Check Point
 - Fact-Checking eBook Content of Halliday Resnick Check Point
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
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