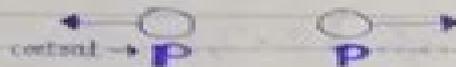


a) **Force due to electron:** As electron is negatively charged and proton is positively charged so, the direction of force on central proton due to electron is "Leftward".

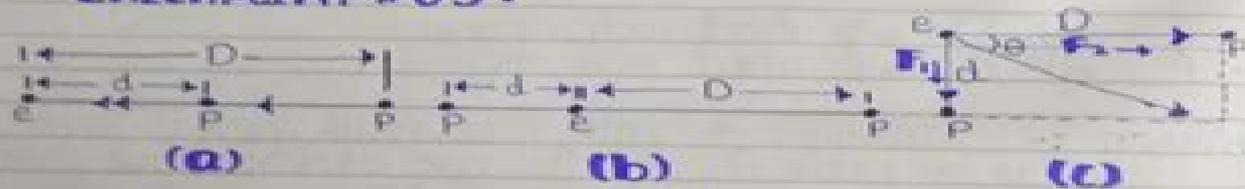


b) **Force due to proton:** As both have same charges so, they will repel each other. The direction of force on central proton due to other proton is "Leftward".



c) **Net Force:** The direction of net force on central proton will be "Leftward" as the force due to electron as well as other proton on central proton is in leftward direction.

CHECKPOINT # 03 :



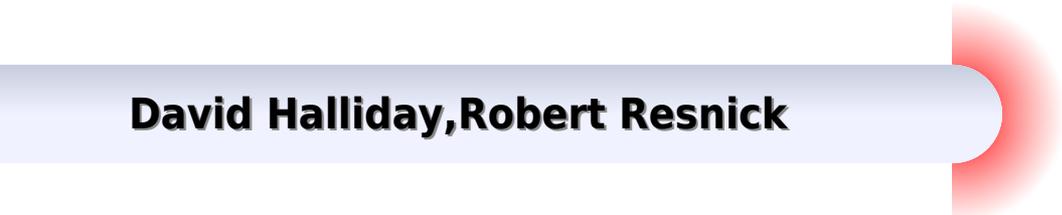
⇒ a) **Ranking Arrangements:** According to the magnitude of the net electrostatic forces on the electron due to protons, the arrangements will be as: **figure (a)** is first as it has largest net electrostatic force as both protons exert attractive force on electron in same direction.

⇒ b) **Angle:** **figure (c)** shows us that $F_1 > F_2$ and
 $\tan \theta = \frac{F_1}{F_2} \rightarrow \theta = \tan^{-1} \frac{F_1}{F_2}$

We can conclude $\theta > 45^\circ$

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