

Chapter 3- Biological Macromolecules

An organic molecule contains carbon, and has C-H bond

An organic molecule contains both carbon and hydrogen. Methane is a simple organic molecule

- Why are all organic molecules (and thus all living things) carbon based?
Carbon= small & versatile

Organic molecules have functional groups

- Functional groups: small groups of non-hydrogen atoms in organic molecules
- They make each organic molecule unique: protein vs. carbohydrate vs....

A monomer is a single unit of a carbohydrate, protein, or nucleic acid. Monomers join to form polymers

During dehydration

Note: carbohydrates contain only C, H, & O in a 1:2:1 ratio

Carbohydrates:

Carbohydrates include simple sugars and polysaccharides

Monosaccharides are the monomers of carbohydrates

Dehydration synthesis binds two monosaccharides, forming a disaccharide. Hydrolysis separates disaccharides into monosaccharides

Polysaccharides are long chains of carbohydrates

Cellulose: structure

Starch: energy

Glycogen: energy

Insulin= hormone from the pancreas

Glucagon= hormone from the pancreas

- Carbohydrates (sugars) and blood sugar levels
- Monosaccharides and disaccharides lead to quick, short increases in blood sugar
- Polysaccharides lead to slow, longer lasting increases or maintenance

Modern Biology Chapter Three Study Guide

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Table of Contents Modern Biology Chapter Three Study Guide

1. Understanding the eBook Modern Biology Chapter Three Study Guide
 - The Rise of Digital Reading Modern Biology Chapter Three Study Guide
 - Advantages of eBooks Over Traditional Books
2. Identifying Modern Biology Chapter Three Study Guide
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in a Modern Biology Chapter Three Study Guide
 - User-Friendly Interface
4. Exploring eBook Recommendations from Modern Biology Chapter Three Study Guide
 - Personalized Recommendations
 - Modern Biology Chapter Three Study Guide User Reviews and Ratings

- Modern Biology Chapter Three Study Guide and Bestseller Lists
- 5. Accessing Modern Biology Chapter Three Study Guide Free and Paid eBooks
 - Modern Biology Chapter Three Study Guide Public Domain eBooks
 - Modern Biology Chapter Three Study Guide eBook Subscription Services
 - Modern Biology Chapter Three Study Guide Budget-Friendly Options
- 6. Navigating Modern Biology Chapter Three Study Guide eBook Formats
 - ePub, PDF, MOBI, and More
 - Modern Biology Chapter Three Study Guide Compatibility with Devices
 - Modern Biology Chapter Three Study Guide Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Modern Biology Chapter Three Study Guide
 - Highlighting and Note-Taking Modern Biology Chapter Three Study Guide
 - Interactive Elements Modern Biology Chapter Three Study Guide
- 8. Staying Engaged with Modern Biology Chapter Three Study Guide
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Modern Biology Chapter Three Study Guide
- 9. Balancing eBooks and Physical Books Modern Biology Chapter Three Study Guide
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Modern Biology Chapter Three Study Guide
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Modern Biology Chapter Three Study Guide
 - Setting Reading Goals Modern Biology Chapter Three Study Guide
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Modern Biology Chapter Three Study Guide
 - Fact-Checking eBook Content of Modern Biology Chapter Three Study Guide
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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