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Chemistry - Unit 7 Review

Chemical Reaction Model

 Describe key characteristics of all chemical reactions, including the role of energy. Explain how a balanced equation represents these features (include an example).

In chemical reactions, atoms of the reactants recombine to form new substances in the products. There is almost always some transfer between the thermal and chemical energy accounts as well as an exchange of thermal energy via heating (Q) between the system and surroundings. The balanced chemical equation represents this process symbolically, showing that the atoms in the products are the same as those in the reactants

 $2 H_2 + O_2 \rightarrow 2 H_2O + energy$

Note: there are 4 atoms of H and 2 atoms of O on each side of the equation.

Applying the Model

Balance the reaction below by drawing any additional reactant and/or product molecules that would be needed.



Write the balanced equation for this reaction.

$$DB_2 + CD_3 + BAC \rightarrow C_2D_4 + AB_3$$

Which of these molecules would be called reactants?

Modeling Chemistry Unit 9 Review Answers

Guillaume Favre

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Cumulated Index Medicus ,1974 The Software Encyclopedia 2000 Bowker Editorial Staff,2000-05

Miscellaneous Publication MP-14: Nuclear Science Teaching Aids and Activities United States Civil and Defense

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,2000-11 EBONY is the flagship magazine of Johnson Publishing Founded in 1945 by John H Johnson it still maintains the highest global circulation of any African American focused magazine Selected Water Resources Abstracts ,1971-10

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Integrated Mathematics Holt McDougal, 1998

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