

## Chemistry – Unit 7 Review

### Chemical Reaction Model

1. Describe key characteristics of all chemical reactions, including the role of energy. Explain how a balanced equation represents these features (include an example).

*In chemical reactions, atoms of the reactants recombine to form new substances in the products. There is almost always some transfer between the thermal and chemical energy accounts as well as an exchange of thermal energy via heating (Q) between the system and surroundings. The balanced chemical equation represents this process symbolically, showing that the atoms in the products are the same as those in the reactants*



*Note: there are 4 atoms of H and 2 atoms of O on each side of the equation.*

### Applying the Model

2. Balance the reaction below by drawing any additional reactant and/or product molecules that would be needed.



Write the **balanced equation** for this reaction.



Which of these molecules would be called reactants?

DB<sub>2</sub>, CD<sub>2</sub>, and BAC are reactants

# Modeling Chemistry Unit 7 Answers

**AN Whitehead**



## **Modeling Chemistry Unit 7 Answers:**

## The Enigmatic Realm of **Modeling Chemistry Unit 7 Answers**: Unleashing the Language is Inner Magic

In a fast-paced digital era where connections and knowledge intertwine, the enigmatic realm of language reveals its inherent magic. Its capacity to stir emotions, ignite contemplation, and catalyze profound transformations is nothing short of extraordinary. Within the captivating pages of **Modeling Chemistry Unit 7 Answers** a literary masterpiece penned by way of a renowned author, readers attempt a transformative journey, unlocking the secrets and untapped potential embedded within each word. In this evaluation, we shall explore the book's core themes, assess its distinct writing style, and delve into its lasting impact on the hearts and minds of those who partake in its reading experience.

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