

Chemistry – Unit 6 Worksheet 1

We have observed evidence that atoms of different elements are different, the most noticeable being in their naturally-occurring chemical behaviour. Atoms of one element cannot be transformed into atoms of another element, which demonstrates that each element's particles are resistant to being transformed. However, when the elements react, each element's particles are transformed to form compounds that do not contain atoms of any one element. We will now examine the patterns that reveal the structure of atoms that allows them to combine in order to describe the changes of the basic building blocks.

1. Write the formula and draw the particle diagram for each compound. The ratio of atoms in each compound is given.

Atomic formula and ratio	12 hydrogen 16 oxygen $\frac{12}{16}$	12 hydrogen 16 oxygen $\frac{12}{16}$	2 hydrogen 1 oxygen $\frac{2}{1}$	2 hydrogen 1 oxygen $\frac{2}{1}$
particle diagram				

Atomic formula and ratio	12 hydrogen 16 oxygen $\frac{12}{16}$	2 hydrogen 1 oxygen $\frac{2}{1}$	2 hydrogen 1 oxygen $\frac{2}{1}$
particle diagram			

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particle diagram				

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RS Peters



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