

# Pre-Ap Chemistry Unit 6 – review for finals

These are the topics you should know and be able to answer questions about:

## 1. Chemical reactions

- What are some examples of evidence of chemical reactions?
  - Evidence that a chemical change *may have* occurred can include:
  - Production of bubbles (gas)
  - Formation of a precipitate (solid)
  - Production of heat and light
  - Color change
  - Noise
  - Ultimately, only the formation of a NEW substance can be absolute proof that a chemical change has occurred.
- What is the difference between chemical and physical reactions?
  - Chemical reactions produce new products – what you start with and end with are different
  - Physical reactions change the appearance of the substance, but you still have the same substance that you start with

## 2. Balance equations.

- Be able to balance equations
  - Either use math or tiddlywinks or draw a picture
- Why do we have to balance equations?
  - Due to the law of conservation of mass. Mass is not created or destroyed during a chemical reaction. Due to this, the mass of the reactants must equal the mass of the products. Which means that there must be a balanced mass on each side of the reaction
- What are we trying to balance when we balance equations?
  - We need the same number of atoms of each element on each side of the arrow
- If you have balanced an equation and notice that the coefficients can be reduced, what do you? And why?
  - Reduce the coefficients.
  - We need the lowest ratio of coefficients
- Draw for me the particle representation of a balanced equation (circles) for the formation of water ( $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O}$ )



- In order to balance the equation  $\text{KClO}_3 \rightarrow \text{KCl} + \text{O}_2$  the correct coefficient for oxygen would be:  
i. 3

## 3. Types of reactions

- What are the different types of reactions?
  - Synthesis
  - Decomposition
  - Single replacement
  - Double replacement
  - Combustion

# Modeling Chemistry Unit 6 Review V1

**Guillaume Favre**



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