

Chemistry – Unit 7 Review

Chemical Reaction Model

1. Describe key characteristics of all chemical reactions, including the role of energy. Explain how a balanced equation represents these features (include an example).

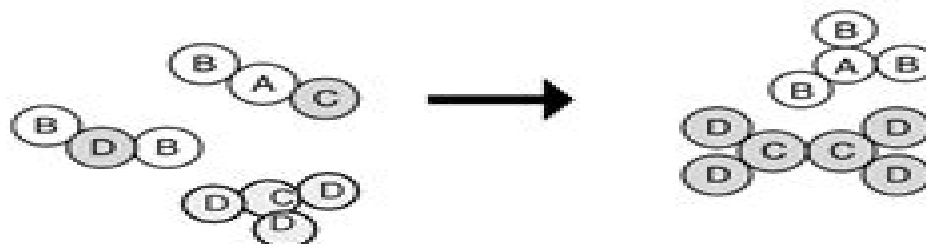
In chemical reactions, atoms of the reactants recombine to form new substances in the products. There is almost always some transfer between the thermal and chemical energy accounts as well as an exchange of thermal energy via heating (Q) between the system and surroundings. The balanced chemical equation represents this process symbolically, showing that the atoms in the products are the same as those in the reactants



Note: there are 4 atoms of H and 2 atoms of O on each side of the equation.

Applying the Model

2. Balance the reaction below by drawing any additional reactant and/or product molecules that would be needed.



Write the **balanced equation** for this reaction.



Which of these molecules would be called reactants?

DB₂, CD₂, and BAC are reactants

Modeling Chemistry Unit 6 Review

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Modeling Chemistry Unit 6 Review:

Chemistry and Industry Review, 1929 **Reviews in Chemistry** Omotayo Ademola Arotiba, Xiao Xiao, Mohammad Mansoob Khan, Jun Chen, Kang Xiao, Tingchao He, 2024-01-03 **Applied Mechanics Reviews**, 1970 *Water Management in Petroleum Industries* Somnath Basu, Andrew R. Shaw, Mudumbai Venkatesh, 2025-08-03 This book provides a roadmap for sustainable development and growth of petroleum industry with respect to water usage and discharge. Water and energy are intricately tied with each other. As a major source of conventional energy, petroleum industries upstream, midstream, and downstream are collectively large consumers of water. Increasing water stress in major parts of the world has made the industry aware of the impact of usable water on different sectors of petroleum industry, e.g., exploration and production, refining, and fuel processing. Treatment of wastewater effluents to maximize reuse is becoming a primary objective of the industry. This, coupled with the need to minimize discharge of contaminants in the effluents that affect human and aquatic life and the environment at large at reasonable cost, is emerging as an important consideration facing the petroleum industry for its sustainable development and growth in the future decades. This book discusses in detail: Sources of water consumed by petroleum production and processing; and wastewater produced; Health and environmental effects of chemicals contained in effluent streams; Effluent treatment processes; current and new innovations and technologies for reuse.

Applied Chemistry and Chemical Engineering, Volume 4 A. K. Haghi, Lionello Pogliani, Eduardo A. Castro, Devrim Balköse, Omari V. Mukbaniani, Chin Hua Chia, 2017-12-22 *Applied Chemistry and Chemical Engineering Volume 4: Experimental Techniques and Methodical Developments* provides a detailed yet easy-to-follow treatment of various techniques useful for characterizing the structure and properties of engineering materials. This timely volume provides an overview of new methods and presents experimental research in applied chemistry using modern approaches. Each chapter describes the principle of the respective method as well as the detailed procedures of experiments with examples of actual applications and then goes on to demonstrate the advantage and disadvantages of each physical technique. Thus, readers will be able to apply the concepts as described in the book to their own experiments. The book is broken into several subsections: Polymer Chemistry and Technology; Computational Approaches; Clinical Chemistry and Bioinformatics; Special Topics. This volume presents research and reviews and information on implementing and sustaining interdisciplinary studies in science, technology, engineering, and mathematics.

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