

Language	1st	2nd/3rd/4th	5th	6th/7th	8th	9th	10th
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Instructions: In Student-Understanding Questions (200 words each)

Directions: Circle the letter that best answers the question or completes the statement. Write the letter that you select in the space provided whenever "Choose ALL that apply" is indicated.

1. Which planet has the strongest magnetic field?

- a. Mars
- b. Neptune
- c. Saturn

- d. Uranus
- e. Venus
- ☒ f. Jupiter

2. Which planet was responsible for the development of the solar system?

- a. Mars
- b. Uranus
- c. Saturn

- d. Neptune
- ☒ e. Jupiter
- f. Venus

3. Which of the following planets shows the evidence of the other in its composition?

- a. Mars, Saturn, Neptune, Uranus
- b. Mars, Uranus, Saturn, Neptune
- ☒ c. Saturn, Neptune, Neptune, Mars

- d. Saturn, Mars, Neptune, Neptune
- e. Neptune, Mars, Neptune, Mars
- f. Neptune, Neptune, Saturn, Mars

4. Which of the following planets is the most massive in the solar system?

- a. Neptune
- b. Saturn
- ☒ c. Jupiter

- d. Neptune
- e. Neptune
- f. Neptune

5. Which of the following planets is the most massive in the solar system?

- a. Saturn
- ☒ b. Jupiter
- c. Saturn

- d. Saturn
- ☒ e. Jupiter
- f. Saturn

6. How many days does it take for the planet to complete one orbit around the Sun?

- a. 1 day
- b. 1 year
- c. 1 day

- ☒ d. 1 year
- e. 1 year
- f. 1 year

Modeling Chemistry Unit 1ws1

Alan Hinchliffe



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Practical Chemistry Unit 1, 1987 **Molecular Modelling for Beginners** Alan Hinchliffe, 2005-12-17 Presenting a concise basic introduction to modelling and computational chemistry this text includes relevant introductory material to ensure greater accessibility to the subject Provides a comprehensive introduction to this evolving and developing field Focuses on MM MC and MD with an entire chapter devoted to QSAR and Discovery Chemistry Includes many real chemical applications combined with worked problems and solutions provided in each chapter Ensures that up to date treatment of a variety of chemical modeling techniques are introduced **Chemical Modeling** Alan Hinchliffe, 1999-10-18 Chemical Modeling From Atoms to Liquids examines materials in terms of the basic properties of atoms molecules and polymer chains In particular the interactions between these fundamental building blocks and the intermolecular and intramolecular potentials are examined Fundamental theories of the constituent particles are covered introducing and developing classical mechanics and quantum mechanics from basics These theories are then applied to modeling developing models from both classical mechanics and quantum mechanics The book aims to make this subject both appealing and relevant whilst avoiding excessive mathematical rigour Carefully structured the text starts by introducing classical quantum and statistical mechanics before moving on to cover the modeling of solids gases and liquids To bring the subject alive many real life examples and applications have been included This book brings together the often scattered and diffuse background information that is essential for a full understanding of chemical modeling FEATURES Assumes no prior knowledge of modeling Brings together all the background information that a reader needs to know to fully understand modeling Develops classical and quantum mechanical theories from basics Avoids unnecessary mathematical rigour Includes a detailed mathematical Toolbox as a ready reference Includes brief descriptions and web addresses of key software packages CONTENTS Introduction Acknowledgements Software Packages Describing Macroscopic Systems Thermodynamics R sum of Classical Mechanics Modeling Simple Solids I Introduction to Quantum Mechanics Electric Multipoles Polarizabilities and Intermolecular Forces Some Statistical Ideas Applications of the Boltzmann Distribution Modeling Simple Solids II Molecular Mechanics Molecular Dynamics and Monte Carlo Techniques The Ideal Monatomic Gas Quantum Gases Introduction to Statistical Thermodynamics Modeling Atoms Diatomics Quantum Modeling of Larger Systems Describing Electron Correlation The Band Theory of Solids Modeling Polymeric Materials Modeling Liquids Appendices Suggestions for Further Reading Index **Introducing Molecular Modeling Into the Undergraduate Chemistry Curriculum** Warren J. Hehre, Janet E. Nelson, 1997 **Molecular Modeling** Hans-Dieter Höltje, 2003 **Using Modeling to Gain a Better Understanding of Chemistry Concepts** Elisabeth Knierim, 2018 *Chemistry* Nicholas Derry, Nick Derry, Lanna Derry, 2000-01 *Programmed Unit in Chemistry* Virginia P. Powell, 1965

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