

## UNIT 4: CHEMICAL REACTIONS STUDY GUIDE

Multiple choice:

\_\_\_**C**\_\_\_ 1. Chemical reactions \_\_\_\_\_.

- a. simply change the form or state of the reactants.
- b. create new atoms.
- c. always produce new substances.
- d. create more matter than what they started with.

\_\_\_**B**\_\_\_ 2. Which of the following is the correct skeleton equation for the combination of solid zinc with liquid bromine to produce solid zinc bromide?

- a.  $\text{Zn}_{2(s)} + \text{Br}_{(l)} \rightarrow \text{ZnBr}_{2(s)}$
- b.  $\text{Zn}_{(s)} + \text{Br}_{2(l)} \rightarrow \text{ZnBr}_{2(s)}$
- c.  $\text{Zn}_{(s)} + \text{Br}_{2(aq)} \rightarrow \text{ZnBr}_{2(s)}$
- d.  $\text{ZnBr}_{2(s)} \rightarrow \text{Zn}_{(s)} + \text{Br}_{2(l)}$

\_\_\_**A**\_\_\_ 3. A substance that is needed to speed up a reaction but is not a product or a reactant is known as a \_\_\_\_\_.

- a. catalyst
- b. coefficient
- c. spectator ion
- d. steroid

\_\_\_**D**\_\_\_ 4. In a combustion reaction, one of the reactants is always \_\_\_\_\_.

- a. a liquid
- b. water
- c. a metal
- d. oxygen

\_\_\_**B**\_\_\_ 5. Which of the following statements is true about the decomposition of a simple binary compound?

- a. The products cannot be predicted.
- b. The reactant is a single substance.
- c. The reactant must always be a molecular substance.
- d. The product is a single substance.

\_\_\_**C**\_\_\_ 6. The reaction  $2\text{P}_2\text{O}_5 \rightarrow 4\text{P} + 5\text{O}_2$  is an example of which type of reaction?

- a. double replacement
- b. single replacement
- c. decomposition
- d. combustion
- e. synthesis

# Modeling Chemistry U4 Test A V3 0

**Lei Huang**



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### **Table of Contents Modeling Chemistry U4 Test A V3 0**

1. Understanding the eBook Modeling Chemistry U4 Test A V3 0
  - The Rise of Digital Reading Modeling Chemistry U4 Test A V3 0
  - Advantages of eBooks Over Traditional Books
2. Identifying Modeling Chemistry U4 Test A V3 0
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Modeling Chemistry U4 Test A V3 0
  - User-Friendly Interface
4. Exploring eBook Recommendations from Modeling Chemistry U4 Test A V3 0
  - Personalized Recommendations

- Modeling Chemistry U4 Test A V3 0 User Reviews and Ratings
- Modeling Chemistry U4 Test A V3 0 and Bestseller Lists
- 5. Accessing Modeling Chemistry U4 Test A V3 0 Free and Paid eBooks
  - Modeling Chemistry U4 Test A V3 0 Public Domain eBooks
  - Modeling Chemistry U4 Test A V3 0 eBook Subscription Services
  - Modeling Chemistry U4 Test A V3 0 Budget-Friendly Options
- 6. Navigating Modeling Chemistry U4 Test A V3 0 eBook Formats
  - ePub, PDF, MOBI, and More
  - Modeling Chemistry U4 Test A V3 0 Compatibility with Devices
  - Modeling Chemistry U4 Test A V3 0 Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Modeling Chemistry U4 Test A V3 0
  - Highlighting and Note-Taking Modeling Chemistry U4 Test A V3 0
  - Interactive Elements Modeling Chemistry U4 Test A V3 0
- 8. Staying Engaged with Modeling Chemistry U4 Test A V3 0
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Modeling Chemistry U4 Test A V3 0
- 9. Balancing eBooks and Physical Books Modeling Chemistry U4 Test A V3 0
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Modeling Chemistry U4 Test A V3 0
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Modeling Chemistry U4 Test A V3 0
  - Setting Reading Goals Modeling Chemistry U4 Test A V3 0
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Modeling Chemistry U4 Test A V3 0
  - Fact-Checking eBook Content of Modeling Chemistry U4 Test A V3 0

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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