

1. Which is not an example of exothermic reaction :



2. Which is an example of endothermic reaction :



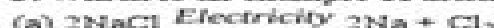
3. Which is an example of combination reaction :



4. Which is an example of decomposition reaction?



5. Which is an example of thermal decomposition?



6. Which is an example of electrolysis decomposition?



7. Which is an example of photo chemical reactions?



8. Which compound is used in photography?



9. Which is an example of double displacement reaction?



10. $\text{Fe}_2\text{O}_3 + 2\text{Al} \rightarrow \text{Al}_2\text{O}_3 + 2\text{Fe}$ is an example of

(a) Combination reaction

(c) Double displacement reaction

11. Which is an example of double displacement reaction?



12. $\text{H}_2\text{S} + \text{Cl}_2 \rightarrow 2\text{HCl} + \text{S}$ is this reaction 'Cl₂' act as :

(a) Oxidising agent (b) Reducing agent

13. Which statement is correct regarding the reaction $2\text{PbO} + \text{C} \rightarrow \text{C}_2\text{Pb} + \text{CO}_2$

(a) Lead is getting reduced (b) Carbon dioxide is getting oxidised

(c) Lead oxide is getting reduced (d) All of these

14. In the reaction $\text{MnO}_2 + 4\text{HCl} \rightarrow \text{MnCl}_2 + 2\text{H}_2\text{O} + \text{Cl}_2$, MnO_2 act as?

(a) Reducing agent (b) Oxidising agent

15. Which is an example of redox reaction?



16. Which is a formula of rust?



17. Rusting of Iron is an example:

(a) Oxidation (b) Reduction

18. Antioxidant is/are:

(a) Oxidising agent (b) Reducing agent

19. Rancidity can be prevented by:

(a) Adding antioxidants



(d) All of these



(d) All of these



(d) All of these



(d) All of these



(d) All of these



(d) (a) and (b)



(d) All of these

(b) Decomposition reaction

(d) Displacement reaction



(c) Both (a) & (b)

(d) None of these

(c) Both (a) & (b)

(d) None of these



(d) All of these



(c) Redox

(d) All of these

(c) Both (a) & (b)

(d) None of these

(b) Storing food in air light

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K Payea



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