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Chemistry - Unit 7 Review

Chemical Reaction Model

 Describe key characteristics of all chemical reactions, including the role of energy. Explain how a balanced equation represents these features (include an example).

In chemical reactions, atoms of the reactants recombine to form new substances in the products. There is almost always some transfer between the thermal and chemical energy accounts as well as an exchange of thermal energy via heating (Q) between the system and surroundings. The balanced chemical equation represents this process symbolically, showing that the atoms in the products are the same as those in the reactants

 $2 H_2 + O_2 \rightarrow 2 H_2O + energy$

Note: there are 4 atoms of H and 2 atoms of O on each side of the equation.

Applying the Model

Balance the reaction below by drawing any additional reactant and/or product molecules that would be needed.



Write the balanced equation for this reaction.

$$DB_2 + CD_3 + BAC \rightarrow C_2D_4 + AB_3$$

Which of these molecules would be called reactants?

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D Kirk

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Energy Research Abstracts ,1982 Reviews in Computational Chemistry, Volume 7 Kenny B. Lipkowitz, Donald B. Boyd,2009-09-22 This is the seventh volume in the successful series designed to help the chemistry community keep current with the many new developments in computational techniques The writing style is refreshingly pedagogical and non mathematical allowing students and researchers access to computational methods outside their immediate area of expertise Each invited author approaches a topic with the aim of helping the reader understand the material solve problems and locate key references quickly Biota and Biological Parameters as Environmental Indicators Phillip E. Greeson,1981

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