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Chemistry – Unit 10 Worksheet 3

1. Element X has two natural isotopes. The isotope with a mass number of 6 has a relative abundance of 7.5%. The isotope with a mass number of 7 has a relative abundance of 92.5%. Determine the average molar mass for the element from these figures. What is the true identity and atomic number of element X?

 The element copper is found to contain the naturally occurring isotopes 29Cu⁶³ and 29Cu⁶⁵. The relative abundances are 69.1% and 30.9% respectively. Calculate the average molar mass of copper.

Uranium has three isotopes with the following percent abundances: 92U²³⁴ (0.0058%), 92U²³⁵(0.71%), 92U²³⁸ (99.23%). What do you expect the molar mass of uranium to be in whole numbers? Why?

4. A sample of silver as it occurs in nature is 52.0% of isotope 47Ag¹⁰⁷ and 48.0% of isotope 47Ag¹⁰⁹. What is the average molar mass of silver? (Compare your result with the value given in the periodic table).

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Technical Abstract Bulletin, Carbon Cycles and Climate Jerry S. Olson, Linda J. Allison, Bobbie-Neal Collier, 1980 This partially annotated bibliography contains the first 1000 references from a computerized file of literature on the global ecological implications of carbon cycles and climatic changes Many early citations originated from the Biogeochemical Ecological Information Center established at Oak Ridge National Laboratory in 1968 and from profiles of computerized files such as Government Research Abstracts GRA and Biological Abstracts BA Later citations have been extracted from the open literature through 1978 and early 1979 from government reports and impact statements and from profiles of GRA BA and the Energy Data Base of the Department of Energy Technical Information Center Oak Ridge Tennessee The subject categories covered by this bibliography may be divided into two main topics carbon cycling and climate system analysis Volume I contains an introduction and overview Volume 2 contains an alphabetical by author listing of citations Volume 3 provides indexes for author organization corporate authority keywords or free index terms taxonomic category subject category Chemical Abstracts codes Biological Abstracts codes crosscode and COSATI Weekly Government Abstracts codes concentrated with permuted title words Molecular Biology of Membrane Transport Disorders Thomas E. Andreoli, A.M. Brown, D.M. Fambrough, Joseph F. Hoffman, Stanley G. Schultz, Michael J. Welsh, 2013-11-11 When the six of us gathered to start planning for what was to be the Third Edition of Physiology of Membrane Disorders it was clear that since 1986 when the Second Edition appeared the field had experienced the dawning of a new era dominated by a change in focus from phenomenology to underlying mechanisms propelled by the power of molecular biology In 1985 detailed molecular information was available for only three membrane transporters the lac permease bacterial rhodopsin and the acetylcholine receptor During the decade that has since elapsed almost all of the major ion channels and transport proteins have been cloned sequenced mutagenized and expressed in homologous as well as heterologous cells Few if any of the transporters that were identified during the previous era have escaped the probings of the new molecular technologies and in many instances considerable insight has been gained into their mechanisms of function in health and disease Indeed in some instances novel

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