## Solutions: Metathesis Reactions and Net Ionic Equations

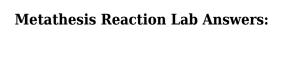
### 9 Pre-lab Questions

Before beginning this experiment in the laboratory, you should be able to answer the following questions.

- Write molecular, complete ionic, and net ionic equations for the reactions that occur, if any, when solutions
  of the following substances are mixed:
  - (a) nitric acid and potassium carbonate
  - (b) sodium bromide and lead nitrate
  - (c) acetic acid and calcium hydroxide
  - (d) calcium nitrate and sodium sulfate
  - (e) ammonium chloride and lithium hydroxide
- 2. Which of the following are not water-soluble: Ba(NO<sub>3</sub>)<sub>2</sub>, FeCl<sub>3</sub>, CuCO<sub>3</sub>, CuSO<sub>4</sub>, ZnS, and ZnSO<sub>4</sub>?
- 3. Write equations for the decomposition of H2CO3 and H2SO3.
- 4. At what temperature (from your graph) do KNO3 and NaCl have the same molar solubility?

## **Metathesis Reaction Lab Answers**

Ralph A. Burns



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