

ঘের ও কীটপতঙ্গের অংশের নির্ধারিত পদ্ধতির প্রয়োগের

চলকের ঘনমাত্রা,

$$S = \frac{W}{V} = \frac{20}{36.5} = 1$$

$$M = 0.5479 \text{ M}$$

অতএব, HCl চলকের ঘনমাত্রা

$$= 0.5479 \text{ M}$$

এখানে,

সংকেত, $m = 20 \text{ g}$

HCl এর আপেক্ষিক ভর,

$$M = 36.5 \text{ g/mol}$$

চলকের আয়তন, $V = 1 \text{ L}$

চলকের ঘনমাত্রা $S = 1$

উদাহরণে উল্লিখিত X-পদ্ধতিতে 10 mg এর মধ্যে কয় পরিমাণ ঘন
র Y-পদ্ধতিতে 0.5 g 2% $\left(\frac{W}{V}\right)$ HCl নমুনা নিয়ে কাজ করা হয়েছে।

যদিও X- পদ্ধতিটি হলো সেমি-মাইক্রো আনালিটিক্যাল পদ্ধতি ও
Y-পদ্ধতিটি হলো ছোটো আনালিটিক্যাল পদ্ধতি। উভয় পদ্ধতির
মধ্যে সেমি-মাইক্রো পদ্ধতি পরিবেশ বান্দব। নিম্নে তা বিশ্লেষণ করা
হলো-

সেমি-মাইক্রো পদ্ধতি রাসায়নিক গুণগত বিশ্লেষণের একটি পদ্ধতি।
এটি বিশ্লেষণের সময় যথেষ্ট রাসায়নিক উপাদান সন্নিবেশন ও
বিশ্লেষণের বিশ্লেষণে ব্যবহৃত হয়। এ পদ্ধতিতে খুব কম পরিমাণ (0.05
থেকে 0.2g কঠিন বা 2-4ml তরল দ্রব্য) রাসায়নিক পদার্থ নিয়ে
কাজ করা হয়। ফলে পরিবেশে মৃদুপ উল্লেখযোগ্যভাবে কমে আসে।
এছাড়া সেমি-মাইক্রো পদ্ধতিতে H_2S এর পরিবর্তে
ক্লোরোফর্মিক অম্ল ব্যবহার করা হয় ফলে পরিবেশে মৃদুপ ঘটে না।



অতএব, কোনো কোনো ক্ষেত্রে কম কষ্টের রাসায়নিক দ্রব্য ব্যবহৃত হয়।
ফলে পরিবেশে মৃদুপের সম্ভাবনা কমে আসে। তাই সেমি-মাইক্রো পদ্ধতি
পরিবেশ বান্দব।

চিত্র-১০



১. উদাহরণে চলক কী?
২. সেমি-মাইক্রো পদ্ধতি পরিবেশ বান্দব কেন?
৩. চলকের জন্য প্রয়োজ্য হ্যাণ্ডলার প্রক্রিয়াটি অঙ্কন করে এবং
ব্যাখ্যা করে।
৪. উদাহরণের অংশের চলক থেকে উপযুক্ত কাচ পাত্র ব্যবহার
করে 1M ঘনমাত্রার চলক তৈরি করার প্রক্রিয়া ব্যাখ্যা করে। ৪
ও না প্রণের উত্তর

৫. বিশ্লেষণের নিমিত্ত কীমের উপর রাইডার স্থাপন করলে কীমের প্রতি
সাপেক্ষে অন্য যে দর পাওয়া যায়, তাকে রাইডার চলক বলে।

৬. সেমি-মাইক্রো পদ্ধতি রাসায়নিক গুণগত বিশ্লেষণের একটি পদ্ধতি।
এটি বিশ্লেষণের সময় যথেষ্ট রাসায়নিক উপাদান সন্নিবেশন ও
বিশ্লেষণের বিশ্লেষণে ব্যবহৃত হয়। এ পদ্ধতিতে খুব কম পরিমাণ
(0.05 থেকে 0.2g কঠিন বা 2-4ml তরল দ্রব্য) রাসায়নিক পদার্থ নিয়ে
কাজ করা হয়। ফলে পরিবেশে মৃদুপ উল্লেখযোগ্যভাবে কমে আসে।
অতএব, কোনো কোনো ক্ষেত্রে কম কষ্টের রাসায়নিক দ্রব্য ব্যবহৃত হয়।
ফলে পরিবেশে মৃদুপের সম্ভাবনা কমে আসে। তাই সেমি-মাইক্রো পদ্ধতি
পরিবেশ বান্দব।

৭. উদাহরণের চলকটি হলো সালফিউরিক এসিড (H_2SO_4) এর 10 M
চল। H_2SO_4 একটি জারক ও ক্ষয়কারী পদার্থ। H_2SO_4 এর 10 M
চল যখন গ্লাস H_2SO_4 চল। গ্লাস H_2SO_4 পদার্থকে ক্ষয়কারী হ্যাণ্ডলার
প্রক্রিয়া অঙ্কন করা হয়। গ্লাস H_2SO_4 এর জন্য ক্ষয়কারী হ্যাণ্ডলার
প্রক্রিয়া অঙ্কন করে ব্যাখ্যা করা হলো:

প্রক্রিয়া	ক্ষয়কারী (Corrosive)
আমদান	ক্ষয়কারী রাসায়নিক উপাদান চলতে যে ক্ষয় রাসায়নিক দ্রব্যকে খুবই দর সম্পর্কে তুলে ও অন্য পদার্থের ক্ষয়কারী ক্ষতি সঞ্চিত হয়। এতে তুলে খুঁচে দর এবং অন্যদর পদার্থ পড়ে যায়।
উদাহরণ	গ্লাস কঠিন এসিডসলভল (HCl , HNO_3 , H_2SO_4), গ্লাস ক্ষয় (NaOH , KOH , H_2O_2), অ্যাসেনিক লিকার প্রক্রিয়া চল ইত্যাদি।
সংরক্ষণ পদ্ধতি	• গ্লাস এসিডকে পোতলে রাখতে হবে। • গ্লাস ক্ষয়কে, ক্ষয় সম্পর্কে বিপাক পোতলে রাখতে হবে।

৮. উদাহরণের চলকটি হলো 10M H_2SO_4 চল এবং এর থেকে 1M
ঘনমাত্রার চলক প্রস্তুত করতে হবে 1ম ও ২ম চলকের ব্যবহার করতে
হবে।

1ম চলকপ্রক্রিয়া হলো 100 ml অ্যাসেনিক লিকার ও এর 10 ml
লিকার। সেমি-অ্যাসেনিক লিকারের অ্যাসেন 100 ml, তাই
আমাদের 100 ml 1M ঘনমাত্রার H_2SO_4 চল প্রস্তুত করতে হবে।
চলকের মৃদুপের খুঁচে অঙ্কন করা হয়।

$$V_1 S_1 = V_2 S_2$$

$$\text{বা, } V_1 = \frac{V_2 S_2}{S_1}$$

$$= \left(\frac{100 \times 1}{10} \right) \text{ ml}$$

$$= 10 \text{ ml}$$

অতএব, 10ml H_2SO_4 চলকের সঙ্গে পানি যোগ করতে হবে
(100-10) ml = 90 ml
অতএব, লিকারের সাথে 10 ml চলকের ঘনমাত্রার চল নিয়ে তার
সঙ্গে অ্যাসেনিক লিকারে 90 ml পানি যোগ করলে আমাদের
কামিষ্ঠ ঘনমাত্রার চল পাওয়া যায়।

চিত্র-১১



১. অ্যাসিডলিকার কীমের চল?
২. সাদা চলক K, এর ঘন কখনো অসীম হয় না-ব্যাখ্যা
করে।
৩. A কীমের চলকে উত্তর করার কৌশল বর্ণনা করে।
৪. B ও C কীমের সাথে সম্পর্কিত রাসায়নিক চলকের
ব্যবহারকালে খাখা কীম দ্রব্যে কবীর উপর সম্পর্কে
বিশ্লেষণ করে। ৪
ও না প্রণের উত্তর

৯. অ্যাসিডলিকার এক ধরনের রাসায়নিক বা কৃত্রিম দ্রব্য সঞ্চিত
বা মৃদুপ খাখা উপাদানসমূহকে জারণ প্রক্রিয়ায় জাল থেকে তুলে
খাখার সঞ্চিত করে।

১০. সামান্য K, এর ঘন কখনো অসীম বা শূন্য হতে পারে না।
উদাহরণ ব্যাখ্যা করার জন্য নিম্নের উদাহরণী মাইক্রোপ্রক্রিয়া
 $A + B \longrightarrow C + D$

Hsc Chemistry 1 First Chapter Notes

L Towne



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