

Practice Final Exam Chem 112

Multiple Choice

Identify the choice that best completes the statement or answers the question.

- _____ 1. Three hundred milliliters (150 mL) of 4.50 M sugar solution is diluted to 600 mL by the addition of pure water. Which of these statements about the solution is correct?
a. The moles of water remain the same. b. The moles of sugar decrease. c. The concentration of sugar increases to 5.0 M. d. The moles of sugar remain constant. e. All of these statements are incorrect.
- _____ 2. How many moles of solute are present in 75.0 grams of solution that is 1.30% by mass glucose ($C_6H_{12}O_6$)? $MM(C_6H_{12}O_6) = 180.16 \text{ g/mol}$
a. 5.4×10^{-3} moles b. 0.800 moles c. 176 moles d. 1.76×10^{-3} moles e. 225 moles
- _____ 3. A concentrated reagent is sold as a solution that is 60.0 mass % solute. If the molar mass of the solute is 244 g/mol and the density of this solution is 1.20 g/mL, what is the molarity of the solution?
a. 10.0 M b. 2.95 M c. 15.5 M d. 0.72 M e. 1.45 M
- _____ 4. The boiling point of pure benzene (C_6H_6) is 80.10 °C. A solution of 14.2 g of a solid dissolved in 64.0 g of benzene boils at 81.09 °C. If k_b for benzene is 2.53 °C/molal, the molar mass of the solid is:
a. 662 g/mol b. 56.0 g/mol c. 92.0 g/mol d. 628 g/mol e. 568 g/mol
- _____ 5. What is the freezing point for a 0.71 m solution of glycerol ($C_3H_8O_3$) in ethanol? Freezing point(ethanol) = -114.6 °C; $K_f(\text{ethanol}) = 1.99 \text{ °C/m}$
a. 0.10 °C b. -113.2 °C c. -1.41 °C d. -116.0 °C e. none of these
- _____ 6. At 450 °C, $K_c = 0.231$ for the reaction:



What is the value of the equilibrium constant for the following reaction at 450 °C?



- a. 0.481 b. 0.156 c. 0.231 d. 0.053 e. 4.329

- _____ 7. Consider the following system at equilibrium and its corresponding concentration equilibrium constant:



What is the value of K_p for this reaction at 470 °C?

- a. $K_p = 3.80 \times 10^{-6}$ b. $K_p = 8.53 \times 10^{-6}$ c. $K_p = 4.32 \times 10^{-5}$ d. $K_p = 1.10 \times 10^{-4}$ e. $K_p = 1.19 \times 10^4$

- _____ 8. For the reaction $\text{I}_2(\text{g}) + \text{Cl}_2(\text{g}) \rightleftharpoons 2 \text{ICl}(\text{g})$, the relation between K_p and K_c is:
a. $K_p = K_c$ b. $K_p = K_c(RT)^2$ c. $K_p = K_c(RT)^{-2}$ d. $K_c = K_p(RT)^2$ e. $K_p = K_c(RT)^4$

Mc Chemistry Final Exam Answers

PT Brinkman



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Physical Chemistry Kenneth S Schmitz, 2018-06-14 The advancements in society are intertwined with the advancements in science To understand how changes in society occurred and will continue to change one has to have a basic understanding of the laws of physics and chemistry Physical Chemistry Multidisciplinary Applications in Society examines how the laws of physics and chemistry physical chemistry explain the dynamic nature of the Universe and events on Earth and how these events affect the evolution of society multidisciplinary applications The ordering of the chapters reflects the natural flow of events in an evolving Universe Philosophy of Science the basis of the view that natural events have natural causes Cosmology the origin of everything from the Big Bang to the current state of the Universe Geoscience the physics and chemistry behind the evolution of the planet Earth from its birth to the present Life Science the molecules and mechanisms of life on Earth Ecology the interdependence of all components within the Ecosphere and the Universe Information Content emphasis on how words and phrases and framing of issues affect opinions reliability of sources and the limitations of knowledge Addresses the four Ws of science Why scientists believe Nature works the way it does Who helped develop the fields of science What theories of natural processes tell us about the nature of Nature and Where our scientific knowledge is taking us into the future Gives a historical review of the evolution of science and the accompanying changes in the philosophy of how science views the nature of the Universe Explores the physics and chemistry of Nature with minimal reliance on mathematics Examines the structure and dynamics of the Universe and our Home Planet Earth Provides a detailed analysis of how humans as members of the Ecosphere have influenced and are continuing to influence the dynamics of events on the paludarium called Earth Presents underlying science of current political issues that shape the future of humankind Emphasizes how words and phrases and framing of issues can influence the opinions of members of society Makes extensive use of metaphors and everyday experiences to illustrate principles in science and social interactions **RBI Grade C (Officer Exam) 15**

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Science Education for the Future Norman Reid, Khaled Ahmed Al-Madani, Fatheya Mahmood Al-Ahmadi, 2025-04-01 This book draws on the wealth of worldwide research into science education to establish a set of key principles for teaching secondary science It considers the aims of science education and the themes and topics that should be included in the curriculum and how these can be effectively taught and assessed in a variety of contexts Offering a framework for the

training and professional development of science teachers the chapters answer key questions such as How can I help my students to make sense of difficult areas in my subject What is meant by understanding and how can I help my students achieve it Is it possible to teach scientific thinking and how can I do it What about learner attitudes when they are sometimes negative What are the different ways to teach in the sciences that are effective How can I make learning in the laboratory more efficient and effective How can I employ assessment as a positive aid to learning Including reflective questions tasks and diagrams this is essential reading for all student and practising secondary science teachers Defending Standardized Testing Richard Phelps, 2005-03-23 Although much has been written on standardized testing policy most of the material has been written by opponents The contributing authors of this volume are both accomplished researchers and practitioners who are respected and admired worldwide They b *Chemistry for the IB Diploma Coursebook with Free Online Material* Steve Owen, Peter Hoeben, Mark Headlee, 2014-03-13 Chemistry for the IB Diploma Second edition covers in full the requirements of the IB syllabus for Chemistry for first examination in 2016 The Second edition of this well received Coursebook is fully updated for the IB Chemistry syllabus for first examination in 2016 comprehensively covering all requirements Get the best coverage of the syllabus with clear assessment statements and links to Theory of Knowledge International mindedness and Nature of Science themes Exam preparation is supported with plenty of sample exam questions online test questions and exam tips Chapters covering the Options and Nature of Science assessment guidance and answers to questions are included in the additional online material available with the book CUET-PG MSc Life Science Practice Set Book 3400+ Question Answer Unit Wise [8 UNits] With Explanations Question Bank DIWAKAR EDUCATION HUB , 2022-08-18 CUET Life Science PGQP22 Complete Practice Question Answer Sets 3400 MCQ Unit Wise from Cover All 8 Units Techniques Chromatin structure and function Biochemistry Biotechnology Microbiology Molecular Genetics Plant Sciences Animal Sciences Highlights of CUET Life Science Question Bank 3400 Questions Answer Included With Explanation 400 MCQ of Each UNit with Explanations As Per Updated Syllabus Include Most Expected MCQ as per Paper Pattern Exam Pattern All Questions Design by Expert Faculties JRF Holder **Advances in Economic Design** Murat R. Sertel, Semih Koray, 2013-04-17 This book Advances in Economic Design celebrates the birth of SED the Society for Economic Design It grew out of SED 2000 the first International Conference of the Society for Economic Design which took place in Istanbul during 23-27 June 2000 While it is not a proceedings or even a selected proceedings volume it nevertheless contains many chapters which derive from papers presented at SED 2000 although they have typically been substantially reworked extended and deepened Of course all the papers published in this book were anonymously refereed As a collection of selected essays the book displays the state of the art across a broad spectrum of theoretical questions and topics and areas of application which economic designers are investigating This characteristic of the book is reflected in its organization into seven parts A Social Choice and Electoral Systems B Buyers and Sellers C Bargaining D Coalitional Stability and Efficiency E Regulating and Organizing

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