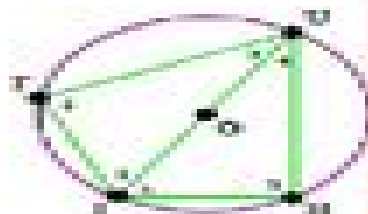


Part Five: Find the measure of the segment, angle or arc for each figure.

29. $m\angle 1$, $m\angle 2$, $m\angle 3$, $m\angle 4$, $m\angle 5$, $m\angle 6$
 when $m\angle 1 = 6x - 30$, $m\angle 3 = 10x$,
 $m\angle 4 = 3y + 21$, $m\angle 6 = 8y + 11$



$$\begin{aligned}\angle 1 + \angle 3 &= 90 \\ 6x - 30 + 10x &= 90 \\ 16x - 30 &= 90 \\ 16x &= 120 \quad x = 7.5\end{aligned}$$

$$\begin{aligned}\angle 5 &= 4(7.5) - 30 = 0 \\ \angle 6 &= 8(7.5) + 11 = 71\end{aligned}$$

$$\begin{aligned}\angle 2 &= 90 \\ \angle 5 &= 0\end{aligned}$$

$$\begin{aligned}\angle 4 + \angle 6 &= 90 \\ 3y + 21 + 8y + 11 &= 90 \\ 11y + 32 &= 90 \quad 11y = 58 \quad y = 5\end{aligned}$$

$$\begin{aligned}\angle 4 &= 3(5) + 21 = 36 \\ \angle 6 &= 8(5) + 11 = 51\end{aligned}$$

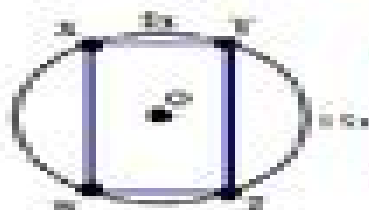
30. $m\angle 1$ and $m\angle 2$



$$\begin{aligned}180 - 90 &= \angle 1 \\ 180 &= \angle 1\end{aligned}$$

$$\begin{aligned}\angle 1 &= \angle 2 = 180 \\ \angle 2 &= 180 = 180 \\ \angle 2 &= 180\end{aligned}$$

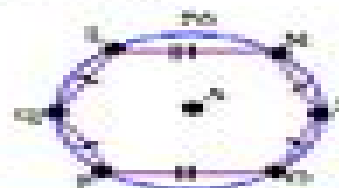
31. $m\widehat{YZ}$



$$\begin{aligned}2(4x) + 2(4x) &= 360 \\ 8x &= 360 \\ x &= 45\end{aligned}$$

$$\begin{aligned}\text{arc } YZ \\ 180 &= 45\end{aligned}$$

32. $m\widehat{MN}$



$$\begin{aligned}2(4x) + 4x &= 360 \\ 8x &= 360 \\ x &= 45\end{aligned}$$

$$\begin{aligned}\text{arc } MN \\ x &= 45\end{aligned}$$

Part Six: The radius of $\odot T$ is 20, $\overline{AB} = 14$ and $m\angle C = 58^\circ$. Find each measure.

33. $m\widehat{AB}$

$$\begin{aligned}\text{arc } AB &= 60 + 60 = 120 \\ (\text{arc addition postulate})\end{aligned}$$

34. \overline{AD}

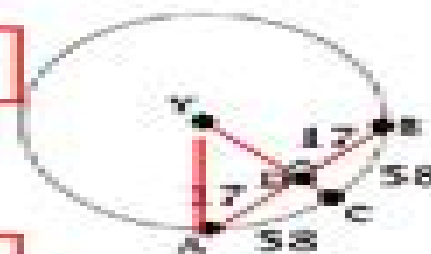
$$\frac{34}{2} = 17 \quad \text{CADO is 1/2 of } \overline{ACD}$$

35. \overline{TD}

$$\begin{aligned}AD^2 + TD^2 &= TA^2 \\ 17^2 + TD^2 &= 20^2 \\ TD &= \sqrt{81} = 9\end{aligned}$$

36. \overline{DC}

$$DC = 20 - 9 = 11$$



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web the volume of 1 00 mol 1 00 mol of any gas at stp standard temperature 273 15 k and pressure 1 atm is measured to be
22 414 l 22 414 l we can substitute 101 325 kpa 101 325 kpa for pressure 22 414 l 22 414 l for volume and 273 15 k 273 15 k
for temperature into the ideal gas equation and solve for r

11 5 charles s law volume and temperature - Feb 19 2022

web may 20 2018 french physicist jacques charles 1746 1823 studied the effect of temperature on the volume of a gas at
constant pressure charles s law states that the volume of a given mass of gas varies directly with the absolute temperature of
the gas when pressure is kept constant the absolute temperature is temperature measured

9 2 relating pressure volume amount and temperature the - Mar 03 2023

web chemists sometimes make comparisons against a standard temperature and pressure stp for reporting properties of
gases 273 15 k and 1 atm 101 325 kpa 1 at stp one mole of an ideal gas has a volume of about 22 4 l this is referred to as the
standard molar volume figure 9 18

lab pressure lab reports lab pressure volume and temperature - Oct 10 2023

web this lab explores the relationships between pressure volume and temperature in an ideal gas the ideal gas law $pV = nRT$ is
used and with it we can explore it by seeing how the pressure and volumes changed in the tubes each time that we

8 2 relating pressure volume amount and temperature the - May 05 2023

web figure pageindex 10 since the number of moles in a given volume of gas varies with pressure and temperature changes
chemists use standard temperature and pressure 273 15 k and 1 atm or 101 325 kpa to report properties of gases

lab report 5 docx pressure volume and temperature - Aug 28 2022

web pressure volume and temperature discovery of gas laws experiment 5 10 3 19 jared caviglia section 582 ian nicholson
introduction the ideal gas law is a commonly used formula in chemistry it relates pressure temperature and volume this
experiment seeks to prove those relationships by measuring the pressure of gas at various temperatures

pressure volume temperature lab ku school of engineering - Jun 25 2022

web the pressure volume temperature pvt fluids lab in ku s chemical petroleum engineering department is used to study the
properties of fluids under a wide range of pressure and temperature

lab report pressure volume and temperature discovery - Jan 01 2023

web volume pressure kpa linear pressure kpa volume ml pressure kpa in part b we observed the relationship between

temperature and pressure the pressure of the 125 ml flask was at 101 36 kpa when the temperature was at 23 2 c

pressure temperature relationship in gases lab studocu - Nov 30 2022

web pressure increases even though there were errors in the experiment the goal of the experiment was met the goal of declaring what type of relationship it is was clear using the data pressure kpa temperature c temperature k constant k p t 95 82 kpa 16 0 c 289 k 0 33 97 90 kpa 19 0 c 292 k 0 34 98 97 kpa 99 22 kpa 101 4 kpa 19 1 c

pressure volume temperature pvt laboratory department - Sep 09 2023

web aug 13 2021 pressure volume temperature pvt laboratory analysis of fluid properties such as api gravity viscosity surface and interfacial tension ph refractive index and vapor pressure is performed in the pvt laboratory pressure volume and temperature relationships of hydrocarbons are studied at the graduate level

che144 thermodynamics engineering lab report saturation vapor pressure - Jul 27 2022

web the properties of water at constant volume can be represented as a function of pressure and temperature as shown in the diagram below figure 1 the pressure temperature relationship at constant volume the saturation point of water is the condition at which a phase change occurs from liquid to vapour or vapour to liquid

6 3 relationships among pressure temperature volume and - Jul 07 2023

web figure pageindex 5 the empirically determined relationships among pressure volume temperature and amount of a gas the thermometer and pressure gauge indicate the temperature and the pressure qualitatively the level in the flask indicates the volume and the number of particles in each flask indicates relative amounts