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Chris Redston & Gillie Cunningham

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web log 10^{-4} 0.5891 $\times 10^{-5}$ mol dm^{-3} log pK_a pH 3.75 4.05 0.30 10.0 30.0 50.0 and $HCOOH$ 0.1 so $HCOO^-$ 0.200 mol dm^{-3} amount vol x

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