



Name: _____

Period: _____

Date: _____

Periodic Table and Periodic Trends Review

1. Why do elements in the same group have similar properties? Same number of valence electrons.
2. Which 3 elements would you assume to have similar properties? O, He, S, Ca, P, Se O, S, Se.
 - Why did you choose those three? They are in the same Group
3. Which elements reside in period 2? Lithium, Beryllium, Boron, Carbon, Nitrogen, Oxygen, Fluorine, Neon.
4. T T or F: Group 1 elements are referred to as alkali metals.
5. F T or F: Calcium is halogen.
6. T T or F: Tungsten is a transition metal.
7. Metals prefer to (gain / lose) electrons and become (cations / anions).
8. Aluminum will (gain / lose) electrons and become a (cation / anion).
9. Chlorine will (gain / lose) electrons and become a (cations / anion).
10. F T or F: Bromine is considered to be a good conductor of electricity.
11. F T or F: Aluminum is a metalloid.
12. T T or F: Silicon is a metalloid.
13. T T or F: Antimony would be considered to have both metal and nonmetal properties.
14. T T or F: Barium has 2 valence electrons.
15. F T or F: Phosphorus has 7 valence electrons
16. Which element has an electron configuration ending in $3p^6$? Argon
17. Which block (s,p,d,f) are the following found?
 - Sodium s
 - Iron d
 - Silicon p
 - Uranium f
18. Which block (s,p,d,f) contains the actinide series? f
19. In general, the electronegativity of an element (increases/decreases) as you move across a period and (increases/decreases) as you move down a group.
20. In general, the atomic radius of an element (increases/decreases) as you move across a period and (increases/decreases) as you move down a group.
21. Identify the period and group of the element with the electron configuration of $[\text{Ne}] 3S^2 3P^4$ Period 3 – Group 16

Elements Periodic Table Trends Review Questions

Daniela Niemeyer



Elements Periodic Table Trends Review Questions:

The Enigmatic Realm of **Elements Periodic Table Trends Review Questions**: Unleashing the Language is Inner Magic

In a fast-paced digital era where connections and knowledge intertwine, the enigmatic realm of language reveals its inherent magic. Its capacity to stir emotions, ignite contemplation, and catalyze profound transformations is nothing in short supply of extraordinary. Within the captivating pages of **Elements Periodic Table Trends Review Questions** a literary masterpiece penned by way of a renowned author, readers attempt a transformative journey, unlocking the secrets and untapped potential embedded within each word. In this evaluation, we shall explore the book's core themes, assess its distinct writing style, and delve into its lasting impact on the hearts and minds of those who partake in its reading experience.

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