# Electron Configuration Practice Worksheet KEY

In the space below, write the unabbreviated electron configurations of the following elements:

- sodium 1s<sup>2</sup> 2s<sup>2</sup> 2p<sup>6</sup> 3s<sup>1</sup>
- 2) iron 1s<sup>2</sup> 2s<sup>2</sup> 2p<sup>6</sup> 3s<sup>2</sup> 3p<sup>6</sup> 4s<sup>2</sup> 3d<sup>6</sup>
- bromine 1s<sup>2</sup> 2s<sup>2</sup> 2p<sup>6</sup> 3s<sup>2</sup> 3p<sup>6</sup> 4s<sup>2</sup> 3d<sup>10</sup> 4p<sup>6</sup>
- 4) barium 1s2 2s2 2p6 3s2 3p6 4s2 3d10 4p6 5s2 4d10 5p6 6s2
- 5) neptunium 1s<sup>2</sup> 2s<sup>2</sup>2p<sup>6</sup>3s<sup>2</sup>3p<sup>6</sup>4s<sup>2</sup> 3d<sup>60</sup> 4p<sup>6</sup> 5s<sup>2</sup> 4d<sup>60</sup> 5p<sup>6</sup> 6s<sup>2</sup> 4f<sup>64</sup> 5d<sup>60</sup> 6p<sup>6</sup> 7s<sup>2</sup> 5f<sup>6</sup>

In the space below, write the abbreviated electron configurations of the following elements:

- 6) cobalt [Ar] 4s<sup>2</sup>3d<sup>7</sup>
- 7) silver [Kr] 5s<sup>2</sup>4d<sup>9</sup>
- 8) tellurium [Kr] 5s24d105p4
- radium [Rn] 7s<sup>2</sup>
- 10) lawrencium [Rn] 7s25f46d1

Determine what elements are denoted by the following electron configurations:

- 11) 1s<sup>2</sup>2s<sup>2</sup>2p<sup>6</sup>3s<sup>2</sup>3p<sup>4</sup> Sulfur
- 12) 1s<sup>2</sup>2s<sup>2</sup>2p<sup>6</sup>3s<sup>2</sup>3p<sup>6</sup>4s<sup>2</sup>3d<sup>10</sup>4p<sup>6</sup>5s<sup>1</sup> Rubidium
- 13) [Kr] 5s<sup>2</sup>4d<sup>10</sup>5p<sup>3</sup> Antimony
- 14) [Xe] 6s<sup>2</sup>4f<sup>14</sup>5d<sup>6</sup> Osmium
- 15) [Rn] 7s25f11 Einsteinium

Determine which of the following electron configurations are not valid:

- 16) 1s<sup>2</sup>2s<sup>2</sup>2p<sup>6</sup>3s<sup>2</sup>3p<sup>6</sup>4s<sup>2</sup>4d<sup>10</sup>4p<sup>5</sup> should be 3d<sup>10</sup>
- 17) 1s<sup>2</sup>2s<sup>2</sup>2p<sup>6</sup>3s<sup>1</sup>3d<sup>5</sup> 3s<sup>3</sup> is impossible& "p" instead of "d"
- 18) [Ra] 7s<sup>2</sup>5f<sup>8</sup> Ra is not a noble gas
- 19) [Kr] 5s<sup>2</sup>4d<sup>10</sup>5p<sup>5</sup> Valid
- 20) [Xe] Element can't be its own electron configuration.

# **Electron Configuration Worksheet And Lots More Key**

**M Lipman** 

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web oct 10 2020 this study aimed to identify the least learned competencies needs and challenges of engineering physics students and teachers and develop a laboratory manual for engineering physics

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web objective of the engineering physics lab manual is to furnish the conceptual understanding of the basic principles the experiments are selected from various areas of physics like lasers fiber optics electricity magnetism and basic electronics the lab manual was written as per the new syllabus

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web each student should bring to the laboratory a physics laboratory notebook coil bound with  $\frac{1}{4}$  or 5 mm grid paper an inexpensive set of geometrical instruments and a calculator with trigonometric functions specific instructions about the materials required for the laboratory will be given in the first meeting of the class absenteeism

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web needs and challenges of engineering physics students and teachers and develop a laboratory manual for engineering physics through a mix method design and addie model but limited to analysis design and development of laboratory manual

only the participants were six engineering physics laboratory teachers and