1. Use noble gases to writ	e the abbreviated electron configurations of t	he following?
i. Li	ii. B	
ш.о	iv. Mg	
v. Cl	vi. V	
vii. Se <sup>2-</sup>	viii. Cr <sup>2+</sup>	
ix. I	x. Ag	
2. Identify the element w	th the following electron configuration	
i. 1s² 2s² 2p° 3s² 3p° 4s² 3d³	'4p4	
ii. 1s² 2s² 2p° 3s² 3p° 4s² 3d	311 4p6 5s2	
iii. 1s² 2s² 2p° 3s² 3p° 4s² 3c	l <sup>10</sup> 4p <sup>6</sup> 5s <sup>1</sup> 4d <sup>7</sup>	
iv. 1s2 2s2 2p4 3s2 3p4 4s2 3c	110 4p° 5s² 4d10 5p° 6s² 4f <sup>14</sup> 5d10	
v. 1s2 2s2 2p4 3s2 3p4 4s2 3d	10 4p <sup>6</sup> 5s <sup>2</sup> 4d <sup>10</sup> 5p <sup>6</sup> 6s <sup>2</sup> 4f <sup>14</sup> 5d <sup>10</sup> 6p <sup>6</sup> 7s <sup>1</sup>	
3. Determine if the follow i. 1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>6</sup> 3s <sup>2</sup> 3p <sup>6</sup> 4s <sup>2</sup> 4d ii. 1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>6</sup> 3s <sup>3</sup> iii. [Rn] 7s <sup>2</sup> 5f <sup>6</sup> 6d <sup>2</sup> iv. [Ar] 5s <sup>2</sup> 4d <sup>10</sup> 5p <sup>3</sup> v. [Xe] 6s <sup>2</sup> 4f <sup>10</sup>	ing electron configurations are correct:  9 4p° 5s¹	
4. Identify the following of	lements	
i. 1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>2</sup>	-	
ii. 1s² 2s² 2pº		
iii. [Ar] 4s <sup>2</sup> 3d <sup>10</sup> 4p <sup>5</sup>		
iv. [Kr] 5s2 4d1		
v. [Ne] 3s2 3p3		

# **Electron Configuration Worksheet And Lots More Answer Sheet**

**Anthony S. Fauci** 

**Electron Configuration Worksheet And Lots More Answer Sheet:** 

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