1 mile = 5,280 ft 1 inch = 2.54 cm 3 feet = 1 yard 454 g = 11b 946 mL = 1 qt

I. Set up and solve the following using dimensional analysis.

1. 5,400 in to mi
2. 16 weeks to sec
3. 54 yards to mm
4. 36 cm/sec to mph
5. 1.09 g/mL to lbs/gal
6. 19 in² to ft²
7. 840 in³ to cm³
8. 4.22 g/cm³ to lbs/ft³
9. 2.50 d/hr to kronin/wk (1 d = 8.60 krc)
10. 32 ft/sec² to meters/min²

Rewrite the following numbers using scientific notation.

1. 476 6. 0.0067 x 10⁻³ 2. 840,000 7. 16 3. 0.0822 8. 0.446 4. 540 x 10³ 9. 28 x 10⁻⁴ 5. 0.000040087 10. 0.0062 x 10⁵

III. How many significant figures are in each of the following numbers or answers to the following mathematical operations.

1. 16.0 6. 14/ 3.07 2. 54,000 7. 5.400 x 10³/ 176 3. 54,000.0 8. 1,874 x 36.2 4. 0.000107 9. 14/ 367 5. 6,007 10. 176/ 1,4809 x 10⁶

IV. Perform the following mathematical operations and express your answers to the proper number of significant figures.

1. 642 x (4.0 x 10⁻⁵) 6. 59 x (3.24 x 10⁻²)/ 4.80 x 10⁴
2. 17/ 3.88 x 10⁷ 7. 42 x (6.02 x 10²³)/ .016
3. (2.9 x 10⁻⁵) x (8.1 x 10²) 8. 12.0/ 6.02 x 10²³
4. (4.3 x 10⁻⁵)³ 9. 0.00000016/ 74.3
5. 5.40 x 10⁻¹⁸/ 769 10. 10.0/ 54,600

 V. Answer the following questions keeping in mind significant figures and dimensional analysis.

1. What is the density of an object that has a mass of 67.0 g and a volume of 14.7 ml.?

2. What is the density of an object that has a mass of 17.0 g and is a cube with dimensions of 1.2 cm x 7.4 cm x 3.0 cm?

3. What volume will 88.0 g of an object with a density of 3.44 g/ mL occupy?

4. How many quarts will 15.0 lbs of a liquid with a density of 2.08 g/ mL occupy?

What will be the mass of 0.047 liters of a substance with a density of 8.73 g/ml.?

Modeling Chemistry Unit 6 5

Emilie Sanchez

Modeling Chemistry Unit 6 5:

Chemistry in the Community American Chemical Society, 2006-01-31 This laboratory based text centres itself around decision making activities where students apply their chemistry knowledge to realistic situations This fifth edition includes more photographs new drawings and new design Applied Chemistry and Chemical Engineering, Volume 4 A. K. Haghi, Lionello Pogliani, Eduardo A. Castro, Devrim Balköse, Omari V. Mukbaniani, Chin Hua Chia, 2017-12-22 Applied Chemistry and Chemical Engineering Volume 4 Experimental Techniques and Methodical Developments provides a detailed yet easy to follow treatment of various techniques useful for characterizing the structure and properties of engineering materials This timely volume provides an overview of new methods and presents experimental research in applied chemistry using modern approaches Each chapter describes the principle of the respective method as well as the detailed procedures of experiments with examples of actual applications and then goes on to demonstrate the advantage and disadvantages of each physical technique Thus readers will be able to apply the concepts as described in the book to their own experiments The book is broken into several subsections Polymer Chemistry and Technology Computational Approaches Clinical Chemistry and Bioinformatics Special Topics This volume presents research and reviews and information on implementing and sustaining interdisciplinary studies in science technology engineering and mathematics **Chemistry in the Community** (ChemCom) American Chemical Society, 2011-06-17 Touted as the most successful NSF funded project published Chemistry in the Community ChemCom by the American Chemical Society ACS offers a meaningful and memorable chemistry program for all levels of high school students ChemCom covers traditional chemistry topics within the context of societal issues and real world scenarios Centered on decision making activities where students are responsible for generating data in an investigating analyzing that data and then applying their chemistry knowledge to solve the presented problem The text is intensively laboratory based with all 39 of the investigations integrated within the text not separate from the reading With the ChemCom program students learn more organic and biochemistry more environmental and industrial chemistry and more on the particulate nature of matter than other textbooks all within the relevance of solving problems that arise in everyday life Meticulously updated to meet the needs of today s teachers and students the new sixth edition of ChemCom adheres to the new science framework as well as the forthcoming next generation of science standards Incorporating advances in learning and cognitive sciences ChemCom's wide ranging coverage builds upon the concepts and principles found in the National Science Education Standards Correlations are available showing how closely aligned ChemCom is to these and other state standards Modelling of Pollutants in Complex Environmental Systems Grady Hanrahan, 2010 Environmental modelling has enjoyed a long tradition but there is a defined need to continually address both the power and the limitations of such models as well as their quantitative assessment This book showcases modern environmental modelling methods the basic theory behind them and their incorporation into complex environmental investigations It highlights

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