

4. a. A sample of a $\text{CaCl}_2 \cdot 2\text{H}_2\text{O} / \text{K}_2\text{C}_2\text{O}_4 \cdot \text{H}_2\text{O}$ solid salt mixture is dissolved in ~150 mL of deionized water previously adjusted to a pH that is basic. The precipitate, after having been filtered, was air-dried and weighed. Data for Trial 1 were obtained as shown. Complete the following table. (See *Report Sheet*.) Record calculated values with the correct number of significant figures.

A. Precipitation of $\text{CaC}_2\text{O}_4 \cdot \text{H}_2\text{O}$ from the Salt Mixture

Calculation Zone

3. Mass of salt mixture (g)	<u>0.879</u>
4. Mass of filter paper (g)	<u>1.896</u>
5. Mass of filter paper and $\text{CaC}_2\text{O}_4 \cdot \text{H}_2\text{O}$ (g)	<u>2.180</u>
6. Mass of air-dried $\text{CaC}_2\text{O}_4 \cdot \text{H}_2\text{O}$ (g)	<u> </u>

Data Analysis, 1.

B. Determination of Limiting Reactant

1. Limiting reactant in salt mixture	<u>$\text{CaCl}_2 \cdot 2\text{H}_2\text{O}$</u>
2. Excess reactant in salt mixture	<u> </u>

Data Analysis, 3.

Data Analysis

1. Moles of $\text{CaC}_2\text{O}_4 \cdot \text{H}_2\text{O}$ precipitated (mol) <i>Show calculation.</i>	<u> </u>
2. Moles of limiting reactant in salt mixture (mol) <i>See equation 8.1.</i>	<u> </u>
3. Mass of limiting reactant in salt mixture (g) <i>Show calculation.</i>	<u> </u>
4. Mass of excess reactant in salt mixture (g) <i>Equals mass of salt mixture minus mass of limiting reactant. Show calculation.</i>	<u> </u>
5. Percent limiting reactant in salt mixture (%) <i>Show calculation.</i>	<u> </u>
6. Percent excess reactant in salt mixture (%)	<u> </u>

Data Analysis, 4.

Data Analysis, 5.

4. b. For Trials 2 and 3, the percent $\text{CaCl}_2 \cdot 2\text{H}_2\text{O}$ in the mixture was 39.7% and 28.4% respectively.

a. What is the average percent of $\text{CaCl}_2 \cdot 2\text{H}_2\text{O}$ in the sample? **Data Analysis, B.**

b. What are the standard deviation and the relative standard deviation (%RSD) for the percent of $\text{CaCl}_2 \cdot 2\text{H}_2\text{O}$ in the sample? **Data Analysis, C and D.**

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