

UNIT 2 CHAPTER STUDY GUIDE

READING AND/OR REVISION

1. **What is matter?**
Matter makes up everything except energy (heat and light)
2. **Give three examples of matter.**
Solid (steel), liquid (water), and gas (air)
3. **Discuss the difference between a physical property and a chemical property.**
Physical properties are observed using five senses without changing the substance. Chemical property is observed when the substance changes.
4. **Give examples of physical properties.**
Color, texture, temperature, shape, size, mass, conductivity, density, magnetism
5. **Give examples of chemical properties.**
Flammability, ability to react with other substances, combustibility, toxicity
6. **Discuss the difference between solubility and a chemical property.**
Solubility is whether a substance can have other things mixed into it. Soluble, and not, is not a property but a state of matter. Type is correct.
7. **What is an element? Give an example of an element.**
It is the smallest amount of matter that cannot be broken down by physical means.
8. **What is an element? Give an example of an element.**
An element cannot be broken down by chemical means. It is made of only one kind of atom.
9. **What is a mixture?**
A mixture is a combination of substances where the substances do not give up their properties. (Homogeneous mixtures)
10. **What is a homogeneous mixture? Give an example of a homogeneous mixture.**
A mixture that looks the same throughout.
11. **What is a heterogeneous mixture? Give an example of a heterogeneous mixture.**
A mixture where you can see the different parts. They have not mixed but are mixed.
12. **What is a solution? Give an example of a solution.**
A homogeneous mixture of two or more substances. The substances are mixed at the molecular level.

What are the states of matter?

1. **What are the three states of matter?**

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