

# **INTERACTIVE SCIENCE 2B** Workbook Solutions (Enrichment Edition)

Chapter 9	SPACE TRAVEL
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## **Part A Sectional Exercise**

### **9.1**

#### **Concept checking**

1. True
2. False
3. True
4. False
5. True
6. False
7. True
8. False

p.1

#### **Question**

$C < B < D < E < A$

p.1

#### **Multiple choice**

1. B
2. A
3. C
4. D
5. B
6. A
7. C
8. E

p.2

### **9.2**

#### **Concept checking**

1. True
2. False
3. False
4. False
5. True
6. True
7. False
8. False

p.3

#### **Fill in the blanks**

1. motion
2. contacting
3. (a) kinetic  
(b) heat
4. air

p.3

# Interactive Science Workbook 2b Answer Second

**Idan Landau**



## **Interactive Science Workbook 2b Answer Second:**

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of laboratory work for understanding science This is evident in more recent books and studies regarding the laboratory which mainly refer to countries with a considerable tradition in practical work in science education Woolnough Alsop 1985 Hodson 1993 Hegarthy Hazel 1990 Wellington 2000 Yet in discussing research studies on labwork several authors express their concern about its effectiveness in facilitating students understanding of various aspects of scientific inquiry They point out a comprehensive re conceptualisation of the aims of labwork and as a consequence of investigating what the students actually learn in different contexts Lazarowitz Tamir 1994 Tobin Tippins 1993 Lunetta 1998 It has also been argued that the relationship between instructional activities and student learning in labwork needs more attention than it has been given in science education research Leach Paulsen 1999 It appears that the case for research based labwork emerges in several quarters in science education particularly among researchers This book presents and discusses a variety of laboratory practices and their effectiveness The studies take into account recent theoretical developments and empirical results concerning students understanding of scientific inquiry A whole chapter is devoted to technological advances offering new learning opportunities for the students and teaching facilities for the teacher

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web 33mol o<sub>2</sub> 6mol br<sub>2</sub> 11mol o<sub>2</sub> 18mol br<sub>2</sub> identify the limiting reactant s and excess reactant s the limiting reactant is o<sub>2</sub> since it would yield the least amount of product 18 mol br<sub>2</sub> the excess reactant is c<sub>2</sub>h<sub>3</sub>br<sub>3</sub> since its complete reaction would have yielded up to 54 mol br<sub>2</sub> think about your result

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web step 4 the reactant that produces a smaller amount of product is the limiting reagent mg mg produces less mgo mgo than does o<sub>2</sub> 0.2398 g mgo vs 25.2 g mgo therefore mg is the limiting reagent in this reaction step 5 the reactant that produces a larger amount of product is the excess reagent

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web if you start with 14.82 g of ca oh<sub>2</sub> c a o h<sub>2</sub> and 16.35 g of h<sub>2</sub>so<sub>4</sub> h<sub>2</sub> s o<sub>4</sub> a determine the limiting reagent b determine the number of moles of h<sub>2</sub>o h<sub>2</sub> o produced c determine the number of grams of caso<sub>4</sub> c a s o<sub>4</sub> produced d determine the number of grams of excess reagent left 1 make sure the equation is balanced

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