

TITRATIONS



Calculate the concentration of the sulfuric acid in mol/dm³

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mean titre = $\frac{22.10 + 22.15 + 22.15}{3} = 22.13 \text{ cm}^3$ ✓

$$n(\text{NaOH}) = 0.105 \times (22.13/1000) = 0.002324 \text{ mol} \quad \checkmark$$

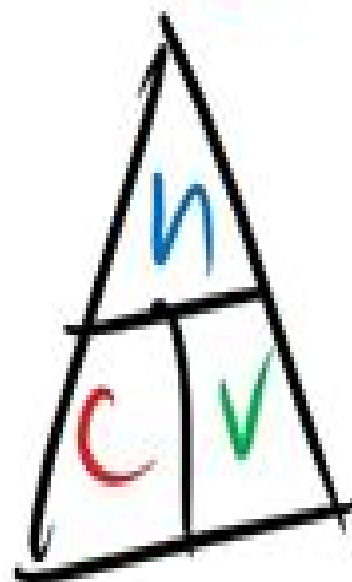
$$n(\text{H}_2\text{SO}_4) = 0.002324 / 2 = 0.001162 \text{ mol} \quad \checkmark$$

$$c(\text{H}_2\text{SO}_4) = \frac{0.001162}{(25/1000)} = 0.0465 \text{ mol/dm}^3 \quad \checkmark$$

$$n = m/M_r$$

$$n = c \times v$$

$$c = n/v$$



GCSE CHEMISTRY

Gcse Titration Practise Questions

**Lesley Butcher, Philip Eastwood, Averil
Macdonald**



Gcse Titration Practise Questions:

Decoding **Gcse Titration Practise Questions**: Revealing the Captivating Potential of Verbal Expression

In a period characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its power to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "**Gcse Titration Practise Questions**," a mesmerizing literary creation penned with a celebrated wordsmith, readers set about an enlightening odyssey, unraveling the intricate significance of language and its enduring affect our lives. In this appraisal, we shall explore the book is central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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