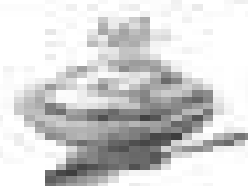


# Atomic Probing: Structure of the Atom

Name \_\_\_\_\_

Date \_\_\_\_\_

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## Purpose:

To determine the structure of the atom by observing the path of alpha particles.

## Part I: Finding Evidence for Nuclear Structure

Alpha particles (helium nuclei) are emitted from a source and pass through a slit in a lead box. They are then detected by a circular screen. The screen is divided into sectors. The sectors are labeled as follows: (a) undeflected, (b) deflected, (c) reflected.

**Observation:** The undeflected sector is the largest, and the deflected and reflected sectors are the smallest.

From this observation, it can be concluded that the atom is mostly empty space.

**Observation:** The deflected sector is the largest, and the undeflected and reflected sectors are the smallest.

From this observation, it can be concluded that the atom has a small, dense nucleus.

**Observation:** The reflected sector is the largest, and the undeflected and deflected sectors are the smallest.

From this observation, it can be concluded that the atom has a small, dense nucleus.

# Lesson 11 Atomic Pudding Answers

**RJ Shavelson**



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