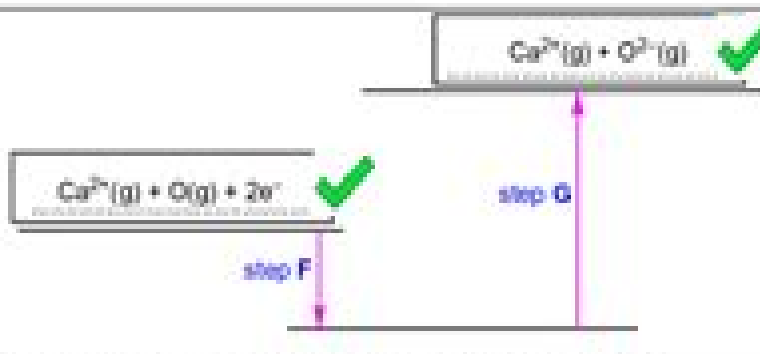


Question	Answer	Marks	Guidance
1 (a)	(The enthalpy change that accompanies) the formation of one mole of a(n ionic) compound ✓ from its gaseous ions (under standard conditions) ✓	2	IGNORE 'energy needed' OR 'energy required' ALLOW as alternative for compound: lattice, crystal, substance, solid Note: 1st mark requires 1 mole 2nd mark requires gaseous ions IF candidate response has '1 mole of gaseous ions', award 2nd mark but NOT 1st mark
(b) (i)		2	Correct species AND state symbols required for both marks $2e^-$ required for left-hand response ALLOW e for e^- Mark each marking point independently
	(ii) (enthalpy change of) formation (of calcium oxide) ✓ (enthalpy change of) atomisation of oxygen ✓ Second electron affinity (of oxygen) ✓	3	calcium oxide not required for this mark DO NOT ALLOW 'lattice formation' (confusion with LE) atomisation AND oxygen/ O_2 / $\frac{1}{2}O_2$ / O both required (atomisation of calcium is also in cycle) IGNORE oxygen or oxygen species, e.g. O^- DO NOT ALLOW calcium

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Table of Contents June 13 Mark Scheme F325

1. Understanding the eBook June 13 Mark Scheme F325
 - The Rise of Digital Reading June 13 Mark Scheme F325
 - Advantages of eBooks Over Traditional Books
2. Identifying June 13 Mark Scheme F325
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an June 13 Mark Scheme F325
 - User-Friendly Interface
4. Exploring eBook Recommendations from June 13 Mark Scheme F325
 - Personalized Recommendations
 - June 13 Mark Scheme F325 User Reviews and Ratings
 - June 13 Mark Scheme F325 and Bestseller Lists

5. Accessing June 13 Mark Scheme F325 Free and Paid eBooks
 - June 13 Mark Scheme F325 Public Domain eBooks
 - June 13 Mark Scheme F325 eBook Subscription Services
 - June 13 Mark Scheme F325 Budget-Friendly Options
6. Navigating June 13 Mark Scheme F325 eBook Formats
 - ePub, PDF, MOBI, and More
 - June 13 Mark Scheme F325 Compatibility with Devices
 - June 13 Mark Scheme F325 Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of June 13 Mark Scheme F325
 - Highlighting and Note-Taking June 13 Mark Scheme F325
 - Interactive Elements June 13 Mark Scheme F325
8. Staying Engaged with June 13 Mark Scheme F325
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers June 13 Mark Scheme F325
9. Balancing eBooks and Physical Books June 13 Mark Scheme F325
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection June 13 Mark Scheme F325
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine June 13 Mark Scheme F325
 - Setting Reading Goals June 13 Mark Scheme F325
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of June 13 Mark Scheme F325
 - Fact-Checking eBook Content of June 13 Mark Scheme F325
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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