

Chapter Test A

Chapter: Chemical Equations and Reactions

on the space provided below the letter of the item or phrase that best completes each statement or best answers each question.

1. The mass number of an atom is the sum of the number of _____ in the nucleus.
 - a. The number of protons and neutrons
 - b. The number of protons and electrons
 - c. The number of electrons in the outer energy level
 - d. None of the above
2. An element described by chemical equation:

$$\text{molecule} + \text{molecule} \rightarrow \text{molecule} + \text{molecule} + \text{molecule}$$
 - a. molecule is produced
 - b. molecule is produced
 - c. molecule is consumed
 - d. None of the above
3. If the chemical equation is balanced, the number of _____ is the same on both sides.
 - a. The number of atoms of each element in the reaction (e.g. number of protons)
 - b. All of the reactants and products
 - c. The number of molecules of each reactant and product
 - d. The number of molecules of each reactant and product
4. There is a chemical reaction if the number of _____ is the same on both sides.
 - a. The number of atoms of each element in the reaction (e.g. number of protons)
 - b. The number of molecules of each reactant and product
 - c. The number of molecules of each reactant and product
 - d. All of the above
5. If the chemical equation is balanced, the number of _____ is the same on both sides.
 - a. The number of atoms of each element in the reaction (e.g. number of protons)
 - b. The number of molecules of each reactant and product
 - c. The number of molecules of each reactant and product
 - d. All of the above

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ML Morrison



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