

7 Ionic and Metallic Bonding

Big idea BONDING AND INTERACTIONS

7.1 Ions

For students using the Foundation edition, assign problems 1, 3–5, 7–12, 14, 15, 18–20.

Essential Understanding Ions form when atoms gain or lose valence electrons, becoming electrically charged.

Lesson Summary

Valence Electrons Valence electrons are the electrons in the outermost occupied energy level and are involved in ion formation.

- ▶ For a representative element, the group number equals the number of valence electrons the atom contains.
- ▶ An electron dot structure shows the symbol of the element and its valence electrons.
- ▶ Atoms tend to gain or lose the number of electrons that will provide the atom with a noble gas electron configuration.

Formation of Cations Cations are positively charged ions formed when an atom loses one or more valence electrons.

- ▶ Atoms and the cations formed from them have different properties.
- ▶ Elements in Group 1 form cations with a charge of $1+$, and those in Group 2 form cations with a charge of $2+$.
- ▶ Many transition metals form more than one cation and do not follow the octet rule.

Formation of Anions Anions are negatively charged ions formed when an atom gains one or more valence electrons.

- ▶ Commonly, the name of an anion ends in *-ide*.
- ▶ Anions form from nonmetallic elements.
- ▶ The anions formed from halogens are known as halides.

After reading Lesson 7.1, answer the following questions.

Valence Electrons

1. What are valence electrons?

Valence electrons are the electrons in the highest occupied energy level of an element's atoms.

2. The valence electrons largely determine the *chemical properties* of an element and are usually the only electrons used in *chemical bonds*.

Lesson 7 Reading And Study Workbook Answers

Paul Krugman, Robin Wells



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Unveiling the Power of Verbal Artistry: An Mental Sojourn through **Lesson 7 Reading And Study Workbook Answers**

In some sort of inundated with screens and the cacophony of instant interaction, the profound energy and mental resonance of verbal art frequently disappear into obscurity, eclipsed by the continuous onslaught of noise and distractions. Yet, set within the lyrical pages of **Lesson 7 Reading And Study Workbook Answers**, a captivating function of literary splendor that impulses with natural emotions, lies an remarkable journey waiting to be embarked upon. Penned with a virtuoso wordsmith, that magical opus courses viewers on an emotional odyssey, softly exposing the latent possible and profound influence stuck within the complex web of language. Within the heart-wrenching expanse of this evocative examination, we shall embark upon an introspective exploration of the book is key styles, dissect its interesting publishing type, and immerse ourselves in the indelible effect it leaves upon the depths of readers souls.

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