1.2 Mole Conversions Solutions

PRACTICE EXAMPLES. To complete the practice examples (1) write the type of question you have, (2) write the equation with units, (3) show the cancellation of units and (4) calculate the solution.

1. Determine how many atoms are present in 2.5 moles of Silicon? (15.05 x 1023 atoms)

MOLES
$$\rightarrow$$
 ATOMS
$$\frac{2.5_moles}{1} \times \frac{6.022 \times 10^{23} atoms}{1_mole} = 15.05 \times 10^{23} atoms$$

2. How many moles of Silicon is 3.01 x 10²³ atoms? (.5 moles)

ATOMS
$$\rightarrow$$
 MOLES
$$\frac{3.011 \times 10^{23} atoms}{1} \times \frac{1 _ mole}{6.022 \times 10^{23} atom} = .5 \text{ moles}$$

3. What is the mass of 4 moles of Iron? (224 grams)

MOLES
$$\rightarrow$$
 GRAMS
$$\frac{4_moles}{1} \times \frac{55.9_grams}{mole} = \frac{224 \text{ grams of C}}{1}$$

4. How many atoms are in 48.6 grams of Magnesium? (12.04 x 1023 atoms)

GRAMS
$$\rightarrow$$
 ATOMS
$$\frac{48.6_grams}{1} \times \frac{1_mole}{24.3_grams} \times \frac{6.022 \times 10^{23} atoms}{1_mole} \approx \frac{12.04 \times 10^{23} atoms}{12.04 \times 10^{23} atoms}$$

5. How many grams are in 18 .06 x 1023 atoms of Copper? (190.5 grams)

ATOMS
$$\rightarrow$$
 GRAMS
$$\frac{18.06 \times 10^{23} atoms}{1} \times \frac{1_mole}{6.022 \times 10^{23} atoms} \times \frac{63.5_grams}{1_mole} \approx 190.5 \text{ grams}$$

How many moles of Nickel is 176 grams? (3 moles)

GRAMS
$$\rightarrow$$
 MOLES
$$\frac{176_grams_of_C}{1} \times \frac{mole}{58.7_grams} \approx 3 \text{ moles}$$

Moles And Formulas Practice Skill Practice 40

O García

Moles And Formulas Practice Skill Practice 40:

The book delves into Moles And Formulas Practice Skill Practice 40. Moles And Formulas Practice Skill Practice 40 is an essential topic that needs to be grasped by everyone, ranging from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Moles And Formulas Practice Skill Practice 40, encompassing both the fundamentals and more intricate discussions.

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