

CHAPTER 11 REVIEW*Molecular Composition of Gases***MIXED REVIEW**

SHORT ANSWER Answer the following questions in the space provided.

1. _____ The average speed of a gas molecule is most directly related to the _____.
(a) polarity of the molecule
(b) pressure of the gas
(c) temperature of the gas
(d) number of moles in the sample
2. _____ Which of the following statements best represents Graham's law for two gases at the same temperature?
(a) The molar masses and the velocities are inversely proportional.
(b) The molar masses and the velocities are directly proportional.
(c) The molar masses and the square roots of the velocities are directly proportional.
(d) The molar masses and the squares of the velocities are inversely proportional.
3. Water is synthesized from its elements according to the following equation:
$$2\text{H}_2(g) + \text{O}_2(g) \rightarrow 2\text{H}_2\text{O}(l)$$
 - _____ a. Which is greater, the volume of H_2 consumed or the volume of O_2 consumed?
 - _____ b. Which is greater, the mass of H_2 consumed or the mass of O_2 consumed?
4. Consider a 2.24 L sample of an ideal gas at STP.
 - _____ a. How many moles are present in the sample, or is more data needed?
 - _____ b. How many molecules are present in the sample, or is more data needed?
 - _____ c. What is the mass of the sample, or is more data needed?

PROBLEMS Write the answer on the line to the left. Show all your work in the space provided.

5. _____ Calculate the volume of 3.5 mol of an ideal gas at 24°C and 760 torr.

Molecular Composition Of Gases Answers

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Fundamentals of Chemistry: A Modern Introduction (1966) Frank Brescia, 2012-12-02 Fundamentals of Chemistry A Modern Introduction focuses on the formulas processes and methodologies used in the study of chemistry The book first looks at general and historical remarks definitions of chemical terms and the classification of matter and states of aggregation The text then discusses gases Ideal gases pressure of a gas confined by a liquid Avogadro's Law and Graham's Law are described The book also discusses aggregated states of matter atoms and molecules chemical equations and arithmetic thermochemistry and chemical periodicity The text also highlights the electronic structures of atoms Quantization of electricity spectra of elements quantization of the energy of an electron associated with nucleus the Rutherford Bohr nuclear theory hydrogen atom and representation of the shapes of atomic orbitals are explained The text also highlights the types of chemical bonds hydrocarbons and their derivatives intermolecular forces solutions and chemical equilibrium The book focuses as well on ionic solutions galvanic cells and acids and bases It also discusses the structure and basicity of hydrides and oxides The reactivity of hydrides charge of dispersal and basicity effect of anionic charge inductive effect and basicity and preparation of acids are described The book is a good source of information for readers wanting to study chemistry GATE Chemistry [CY] Question Bank Book 2800+ Question With Explanation As Per Updated Syllabus DIWAKAR EDUCATION HUB, 2023-09-04 GATE Chemistry Code CY Practice Sets 2800 Question Answer MCQ NAT MSQ Highlights of Question Answer Covered All 21 Chapters Subjects Based MCQ NAT MSQ As Per Syllabus In Each Chapter Unit Given 135 MCQ NAT MSQ In Each Unit You Will Get 135 Question Answer Based on Multiple Choice Questions MCQs Numerical Answer Type NAT Written Type Questions Total 2800 Questions Answer with Explanation Design by Professor JRF Qualified Faculties **Self-Help to ICSE Sample Question Papers Class 10 Chemistry (For 2021 Examinations)**

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Foundations of College Chemistry Morris Hein, Susan Arena, Cary Willard, 2023 Foundations of College Chemistry 16th edition presents chemistry as a modern vital subject and is designed to make introductory chemistry accessible to all beginning students It is intended for students who have never taken a chemistry course or those who had a significant interruption in their studies but plan to continue with the general chemistry sequence The central focus is to make chemistry interesting and understandable and teach students the problem solving skills they will need This International Adaptation offers new and updated content with improved presentation of all course material It builds on the strengths of previous editions including clear explanations and step by step problem solving The material emphasizes real world applications of chemistry as the authors develop the principles that form the foundation for the further study of chemistry There is new and expanded coverage of polarizing power and polarizability Fajans rules collision number and mean free path abnormal molecular masses and van't Hoff factor and applications of radioactivity **American Druggist and**

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- 1 The Mathematics section includes topics like Number System BODMAS Decimals Fractions and Time and Work
- 2 4 General Intelligence and Reasoning will test your skills in analogies coding and decoding and syllogism
- 4 5 The General Science portion will cover Physics Chemistry and Life Sciences at the 10th grade level

1 Exam Pattern The selection process for the RRB Technician includes a Computer Based Test CBT followed by Document Verification and a Medical Examination

- 2 3 The CBT for Technician Grade III will consist of 100 objective type multiple choice questions with a total duration of 90 minutes
- 1 4 There will be a negative marking of 1/3rd of the marks for each incorrect answer
- 3 The minimum qualifying marks are 40% for UR EWS 30% for OBC NCL SC and 25% for ST candidates
- 1 3 Qualification and Eligibility To be eligible for the Technician Grade III posts candidates must have passed Matriculation 10th class along with an ITI certificate in the relevant trade from a recognized institution NCVT SCVT or have completed an Act Apprenticeship
- 6 7 The age limit for applicants is between 18 and 30 years
- 6 8 Number of Vacancies The RRB has announced a total of 6238 vacancies for Technicians out of which a significant number are for the Grade III positions
- 9 10 Specifically there are 6055 posts for Technician Grade III

6 How to Apply Interested and eligible candidates can apply for the RRB Technician recruitment online through the official RRB websites

- 7 11 The application window opened on June 28 2025 and the last date to apply has been extended to August 7 2025
- 7 10 To apply you will need to register on the portal fill out the application form with the required details upload scanned documents and pay the application fee
- 6 Grade Pay and Salary The salary for an RRB Technician Grade III is based on the 7th Pay Commission guidelines
- 12 The initial basic pay for a Technician Grade III is 19 900
- 12 13 In addition to the basic pay employees are also entitled to various allowances such as Dearness Allowance DA House Rent Allowance HRA and Travel Allowance TA which significantly enhances the in hand salary
- 14 15 The grade pay for a Technician Grade III is 1900

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