

Chemistry 530 Midterm Review

Exam on Monday Dec 16 at 8:30

Section I = 75 pts

- Consists of 75 multiple choice questions worth 1 pt. each
- Questions cover all material covered during the second semester
- Questions will be similar to multiple choice problems in the book and review guides on-line
- Practicing the review guides is the best approach; answers will be posted on-line

Section II = 95 pts

- Consists of 19 short answer problems totaling 95 points
- Questions cover the whole semester
- Examples of types of questions
 - Metric Conversions
 - Volume Calculations
 - Percent Composition
 - Atomic Mass
 - Electron configuration
 - Nomenclature
 - Balancing Equations
 - Predicting the products of a reactions
 - Identifying types of reactions
 - Mole Conversions
 - Molar Mass of a compound
 - Molar Mass of a gas at STP
 - Density of a gas at STP
 - Empirical & Molecular Formula

Study approach

- First, read through your notes. Once thoroughly and once more quickly.
- Go through the review packet and make sure you are comfortable with all of the terms and how to do problems
- Do the review packets for help with multiple choice
- Do questions on old tests for practicing written problems
- Make sure you have practiced AT LEAST one of the types of example problems

Chapter 1 – Introduction to Chemistry

- Chemistry: the branch of science that studies the composition and properties of matter.
- Scientific Method:
 - Observation
 - Hypothesis
 - Controlled Experiment
- Theory versus Natural Law

Prerequisite Science Skills

- Rules for Significant Figures:
 - 1) All non-zero digits are significant.
 - 2) Zeros are significant only when they are *between* two non-zero digits or if they occur *after* a decimal point and *after* a non-zero digit.
 - 3) In *multiplication or division*, you cannot end up with more significant figures than you started with.
 - 4) In *addition or subtraction*, you cannot have more digits to the *right* of the decimal than you started with.
- Exponential numbers
- Scientific Notation = $D.DD \times 10^n$ where $1.00 \leq D.DD \leq 9.99$

Modern Chemistry Midterm Study Guide 100 Questions

RJ Shavelson



Modern Chemistry Midterm Study Guide 100 Questions:

Reviewing **Modern Chemistry Midterm Study Guide 100 Questions**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is actually astonishing. Within the pages of "**Modern Chemistry Midterm Study Guide 100 Questions**," an enthralling opus penned by a highly acclaimed wordsmith, readers embark on an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve in to the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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