

balance nuclear reaction equations

aluminum-26 undergoes positron emission
potassium-40 undergoes beta decay (electron)
uranium-238 undergoes alpha decay



electron configurations

Write the electron configuration for C.

Write the electron configuration for Cr.

Write the shorthand notation for Bi.



NOMENCLATURE [naming compounds, writing formulas]

Write the formula for each of the following compounds:

- a. Sulfur hexafluoride SF_6
- b. Lithium nitride Li_3N
- c. Chromium (III) carbonate $\text{Cr}_2(\text{CO}_3)_3$
- d. Tin (II) chloride SnCl_2
- e. Ammonium acetate $\text{NH}_4\text{C}_2\text{H}_3\text{O}_2$ (or $\text{CH}_3\text{COONH}_4$)
- f. Mercury (I) chloride Hg_2Cl_2
- g. Potassium bromate KBrO_3
- h. Hydrosulfuric acid $\text{H}_2\text{S(aq)}$
- i. Chloric acid HClO_3
- j. Sulfurous acid H_2SO_3

Name each of the following compounds.

- a. CuSO_4 Copper (II) sulfate
- b. AlF_3 aluminum fluoride
- c. HI hydrogen iodide
- d. NO nitrogen monoxide
- e. $\text{H}_2\text{Se(aq)}$ hydroselenic acid
- f. HNO_3 nitric acid
- g. HNO_2 nitrous acid
- h. NaHSO_4 sodium hydrogen sulfate

Half-life

Tritium (H-3) is a radioactive isotope of hydrogen with a half-life of 12.3 years. How long would it take for a 40.0 g sample to decay down to 1.25 g?

$$40 \xrightarrow{1} 20 \xrightarrow{2} 10 \xrightarrow{3} 5 \xrightarrow{4} 2.5 \xrightarrow{5} 1.25$$

Fe-61 has a half-life of 6.00 min. Of a 100.0 mg sample, how much will remain after 18.0 min?

$$3 \text{ half lives} \quad 100 \xrightarrow{1} 50 \xrightarrow{2} 25 \xrightarrow{3} 12.5 \text{ mg}$$

Average atomic mass

The element copper has naturally occurring isotopes with mass numbers of 63 and 65.

The relative abundance and atomic masses are 69.2% for a mass of 63amu and 30.8% for a mass of 65amu. Calculate the average atomic mass of copper.

$$\begin{array}{rcl} 63 \times .692 & = & 43.596 \rightarrow 44 \\ 65 \times .308 & = & 20.02 \rightarrow 20 \\ + & & \\ \hline & & 63.62 \rightarrow 64 \end{array}$$

Modern Chemistry Chapter Work Answers

Marcel A. Müller



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Answer Book National Geographic Society (U.S.),2009 Covers everything from earth sciences to astronomy from climate and habitats to human arts and cultures from ancient history to cutting edge technology and descriptions flags and statistics of all the countries in the world *National Geographic Answer Book* National Geographic,2015 This far reaching reference is designed with many entry points and a visually engaging format to satisfy the curious browser the student researcher and the earnest knowledge seeker alike **Electrochemistry** Christine Lefrou,Pierre Fabry,Jean-Claude Poignet,2012-05-24 This textbook offers original and new approaches to the teaching of electrochemical concepts principles and applications Throughout the text the authors provide a balanced coverage of the thermodynamic and kinetic processes at the heart of electrochemical systems The first half of the book outlines fundamental concepts appropriate to undergraduate students and the second half gives an in depth account of electrochemical systems suitable for experienced scientists and course lecturers Concepts are clearly explained and mathematical treatments are kept to a minimum or reported in appendices This book features Questions and answers for self assessment Basic and advanced level numerical descriptions Illustrated electrochemistry applications This book is accessible to both novice and experienced electrochemists and supports a deep understanding of the fundamental principles and laws of electrochemistry Principles of Modern Chemistry David W. Oxtoby,H. Pat Gillis,Laurie J. Butler,2016-01-01 Long considered the standard for honors and high level mainstream general chemistry courses PRINCIPLES OF MODERN CHEMISTRY continues to set the standard as the most modern rigorous and chemically and mathematically accurate text on the market This authoritative text features an atoms first approach and thoroughly revised chapters on Quantum Mechanics and Molecular Structure Chapter 6 Electrochemistry Chapter 17 and Molecular Spectroscopy and Photochemistry Chapter 20 In addition the text utilizes mathematically accurate and artistic atomic and molecular orbital art and is student friendly without compromising its rigor End of chapter study aids focus on only the most important key objectives equations and concepts making it easier for students to locate chapter content while applications to a wide range of disciplines such as biology chemical engineering biochemistry and medicine deepen students understanding of the relevance of chemistry beyond the classroom A Workbook of Electrochemistry John Bockris,2012-12-06 In this book the objective has been to set down a number of questions largely numerical problems to help the student of electrochemical science No collection of problems in electrochemistry has previously been published The challenge which faces the authors of such a book is the breadth of the material in modern electrochemistry and the diversity of backgrounds and needs of people who may find a problems book in electrochemistry to be of use The general intention for Chapters 2 11 has been to give the first ten questions at a level which can be dealt with by students who are undergoing instruction in the science of electrochemistry but have not yet reached graduate standard in it The last two questions in Chapters 2 11 have been chosen at a more advanced standard corresponding to that expected of someone with knowledge at

the level of a Ph D degree in electrochemistry **Volume 1: Modern Electrochemistry** John O'M. Bockris, Amulya K.N. Reddy, 2007-05-08 This book had its nucleus in some lectures given by one of us J O M B in a course on electrochemistry to students of energy conversion at the University of Pennsylvania It was there that he met a number of people trained in chemistry physics biology metallurgy and materials science all of whom wanted to know something about electrochemistry The concept of writing a book about electrochemistry which could be understood by people with very varied backgrounds was thereby engendered The lectures were recorded and written up by Dr Klaus Muller as a 293 page manuscript At a later stage A K N R joined the effort it was decided to make a fresh start and to write a much more comprehensive text Of methods for direct energy conversion the electrochemical one is the most advanced and seems the most likely to become of considerable practical importance Thus conversion to electrochemically powered transportation systems appears to be an important step by means of which the difficulties of air pollution and the effects of an increasing concentration in the atmosphere of carbon dioxide may be met Corrosion is recognized as having an electrochemical basis The synthesis of nylon now contains an important electrochemical stage Some central biological mechanisms have been shown to take place by means of electrochemical reactions A number of American organizations have recently recommended greatly increased activity in training and research in electrochemistry at universities in the United States **Electrical Engineering** ,1916

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Introduction to Physics

in Modern Medicine Suzanne Amador Kane, Boris A. Gelman, 2020-02-13 From x rays to lasers to magnetic resonance imaging developments in basic physics research have been transformed into medical technologies for imaging surgery and therapy at an ever accelerating pace Physics has joined with genetics and molecular biology to define much of what is modern in modern medicine and allied health Covering a wide range of applications Introduction to Physics in Modern Medicine Third Edition builds further on the bestselling second edition Based on the courses taught by the authors the book provides medical personnel and students with an exploration of the physics related applications found in state of the art medical centers Requiring no previous acquaintance with physics biology or chemistry and keeping mathematics to a minimum the application dedicated chapters adhere to simple and self contained qualitative explanations that make use of examples illustrations clinical applications sample calculations and exercises With an enhanced emphasis on digital imaging and computers in medicine the text gives readers a fundamental understanding of the practical application of each concept and the basic science behind it This book provides medical students with an excellent introduction to how physics is applied in medicine while also providing students in physics with an introduction to medical physics Each chapter includes worked examples and a complete list of problems and questions That so much of the technology discussed in this book was the stuff of dreams just a few years ago makes this book as fascinating as it is practical both for those in medicine as well as those in physics who might one day discover that the project they are working on is the basis for the next great medical application Features Introduces state of the art and emerging medical technologies such as optical coherence tomography x ray phase contrast imaging and ultrasound mediated drug delivery Covers hybrid scanners for cancer imaging and the interplay of molecular medicine with MRI CT and PET in addition to intensity modulated radiation therapy and new forms of cancer treatments such as proton and heavy ion therapies Offers an enhanced emphasis on digital imaging and dosimetry including recent innovations in the pixel array x ray detectors ultrasound matrix transducers and direct ion storage dosimeters

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